

1.

Stern-Gerlach magnet

The deflecting force is

$$F = \mu_z \frac{dB_z}{dz},$$

where

$$\mu_z = g_s \mu_B m_s,$$

since $\ell=0$. If D is the deflection and F is constant

$$D = \frac{1}{2} at^2 = \frac{1}{2} \frac{F}{m_e} \left(\frac{L}{v}\right)^2$$

where, L is the magnet length
and v the speed of atoms.

Therefore,

$$D = \frac{1}{2} \frac{(g_s \mu_B m_s)}{m_e} \frac{dB_z}{dz} \left(\frac{L}{v}\right)^2$$

so that

$$\frac{dB_z}{dz} = \frac{2m_e D v^2}{L^2 g_s \mu_B m_s}$$

For atoms emitted from the oven

$$\frac{1}{2} m_e v^2 = 2k_B T$$

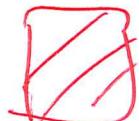
with $T = 1233\text{ K}$.

Hence,

$$\frac{dB_z}{dz} = \frac{8k_B T D}{L^2 g_s \mu_B m_s}$$

$$= \frac{8(1.38 \times 10^{-23})(1233)(0.0005)}{(0.5)^2 2 \times 9.27 \times 10^{-24} \times \frac{1}{2}}$$

$$= 29 \text{ T/m}$$



2.

Hydrogen atom in magnetic field

(a) The orbit and spin energies are $(g_e \mu_B m_e)B$ and $(g_s \mu_B m_s)B$

Hence, with respect to the energy for $B=0$,

$$\Delta E = g_e \mu_B m_e B + g_s \mu_B m_s B$$

$$= (g_e m_e + g_s m_s) \mu_B B = (m_e + 2m_s) \mu_B B$$

(b) For $n=2, l=0, 1$

| l | m_l | m_s | ΔE (units of $\mu_B B$) |
|-----|-------|----------------------------------------------------------------------------|----------------------------------|
| 0 | 0 | $\left\{ \begin{array}{l} \frac{1}{2} \\ -\frac{1}{2} \end{array} \right.$ | +1 |
| | -1 | $\left\{ \begin{array}{l} \frac{1}{2} \\ -\frac{1}{2} \end{array} \right.$ | -1 |
| 1 | -1 | $\left\{ \begin{array}{l} \frac{1}{2} \\ -\frac{1}{2} \end{array} \right.$ | -2 |
| | 0 | $\left\{ \begin{array}{l} \frac{1}{2} \\ -\frac{1}{2} \end{array} \right.$ | +1 |
| | +1 | $\left\{ \begin{array}{l} \frac{1}{2} \\ -\frac{1}{2} \end{array} \right.$ | +2 |

| | l | m_l | m_s | E |
|-------|-----|-------|-------|---------------------|
| | — | 1 | 1 | $\frac{1}{2}$ +2 |
| $n=2$ | = | 0 | 0 | $\frac{1}{2}$ } +1 |
| | | 1 | 0 | $\frac{1}{2}$ } |
| $B=0$ | = | 1 | -1 | $\frac{1}{2}$ } |
| | | 1 | 1 | $-\frac{1}{2}$ } |
| | = | 0 | 0 | $-\frac{1}{2}$ } -1 |
| | | 1 | 0 | $-\frac{1}{2}$ } |
| | — | 1 | -1 | $-\frac{1}{2}$ -2 |

⊗ two-fold degeneracy

§ The maximum separation is

$$\Delta E_{\max} = 4\mu_B B = 10.2 \text{ eV}$$

or

$$4 \times (0.22 \times 10^{-24}) B = (10.2)(1.6 \times 10^{-19})$$

or

$$B = 4.9 \times 10^4 \text{ T}$$

B

1.

The ground state is spatially symmetric, so it goes with the symmetric (triplet) spin configuration.

Thus the ground state is orthohelium and it is triply degenerate.

The excited states come in ortho and para (singlet) form; since the former go with the symmetric spatial wave function, the orthohelium states are higher in energy than the corresponding (nondegenerate)

para states.

(b) The ground state and all excited states come ~~exist~~ in both ortho and para form.

All are quadruply degenerate (or at any rate we have no way a priori of knowing whether ortho or para are higher in energy, since we don't know which goes with the symmetric spatial configuration.)

BZ

3.

(a) Largest

$$j = 4 + \frac{1}{2} = \frac{9}{2}$$

Largest

$$m_j = j = \frac{9}{2}$$

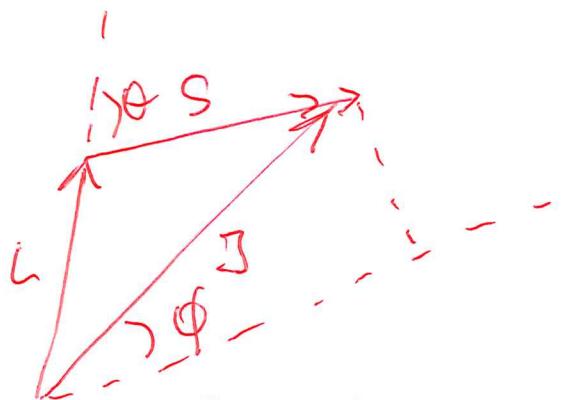
The magnitudes of the vectors are

$$J = \sqrt{j(j+1)} h = \sqrt{9 \cdot 10} h/2$$

$$L = \sqrt{l(l+1)} h = \sqrt{20} h$$

$$S = \sqrt{s(s+1)} h = \sqrt{3} h/2$$

(b)



Apply the law
of cosines to
 L, S, J triangle

$$J^2 = L^2 + S^2 - 2LS \cos(180 - \theta)$$

or

$$\frac{99}{4} = 20 + \frac{3}{q} + 2\sqrt{20 \times \frac{3}{q}^2} \cos \theta$$

so

$$\theta = \cos^{-1}\left(\frac{2}{\sqrt{15}}\right) = 58.91^\circ$$

Since μ_e is antiparallel to L and μ_s is antiparallel to S , the angle between μ_e and μ_s is 58.91°

(c)

$$\cos \phi = \frac{J_z}{J} = \frac{q}{\sqrt{99}}$$

or

$$\phi = 25.24^\circ$$

