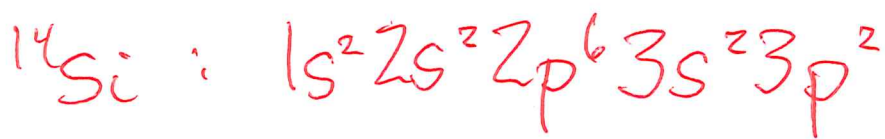
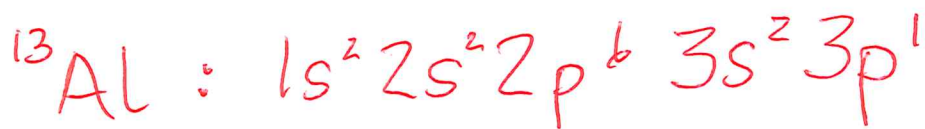
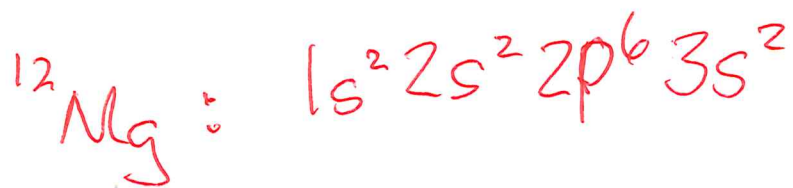


① Ground state configuration

(a) The columns reveal the last shell being filled, the row the numbers of electrons in that shell,

Therefore



(b)

${}^{12}\text{Mg}$: the configuration represents a filled shell, and thus all the angular momenta are zero, 1S_0

^{13}Al : there is a single valence electron ($s = s^1 = 1/2$), thus $2s^1 + 1 = 2$, $l = 1$ giving a P state $J = 3/2, 1/2$ with the smaller J lying lower, $^2P_{1/2}$

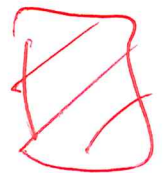
^{14}Si : here there are two $l = 1$ electrons $S = 0, 1$ and $L = 2, 1, 0$. For the lower energy pick the larger S . This gives the possibilities

$L = 2, J = 3, 2, 1, \quad ^3P_{3,2,1}$

$L = 1, J = 2, 1, 0, \quad ^3P_{2,1,0}$

$L = 0, J = 1 \quad ^3S_1$

The 3D_1 and 3S_1 states are however, prohibited by the exclusion principle. Of the ${}^3P_{2,1,0}$ states, the smallest J lies lowest, hence the ground state configuration should be 3P_0 .



2.

Segulvægi: 3P_0

Þetta ástand hefur
heildar hverfipunga $J=0$
þess vegna er segul-
vægið

$$M = g\mu_B \sqrt{J(J+1)} = 0$$

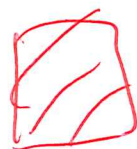


3.

Hverfipungi 3F_4

Stafurinn F segir að heildar brenntar hverfipunga slammtatala h sé 3.

Hástöfunin 3 er margfeldni líðans $2St_1$, sem kemur til vegna spuna slammtafólunnar $S=1$, og á skör eru 4 sem er heildar hverfipunga-slammtatalan $\# 3$.



2

Assign quantum numbers.

multiplet 1:2:3:4

For a single multiplet S and L have the same value for each level.

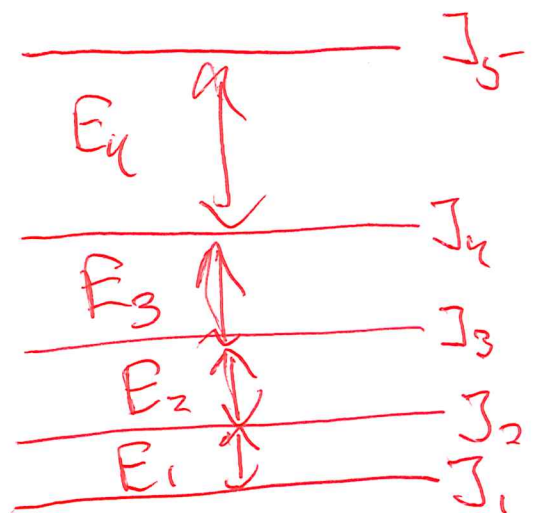
By the interval rule

$$E_4 = 2KJ_5$$

$$E_3 = 2KJ_4$$

$$E_2 = 2KJ_3$$

$$E_1 = 2KJ_2$$



Therefore

$$\frac{E_4}{E_3} = \frac{4}{3} = \frac{J_5}{J_4} = \frac{J_4 + 1}{J_4} \Rightarrow J_4 = 3$$

Since

$$J = 1, J_5 = 4, J_4 = 3, J_2 = 1$$

$$\text{and } J_1 = 0$$

But

$$J = L + S, L + S + 1, \dots, |L - S|$$

so that

$$L + S = 4, L - S = 0$$

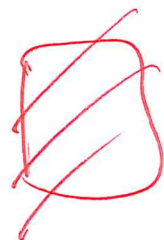
and

$$L = S = 2$$

and hence

$$L = S = 2$$

$$J = 4, 3, 2, 1, 0$$



3. The Zeeman effect

In the absence of magnetic field, the 3d to 2p energy difference is

$$E = (-13.6057 \text{ eV}) \left(\frac{1}{3^2} - \frac{1}{2^2} \right) = 1.88968 \text{ eV}$$

and the wavelength is

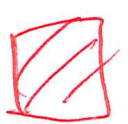
$$\lambda = \frac{hc}{E} = \frac{1239.842 \text{ eV nm}}{1.88968 \text{ eV}} = 656.112 \text{ nm}$$

The magnetic field gives a change in wavelength of

$$\Delta\lambda = \frac{\lambda^2}{hc} \Delta E = \frac{(656.112 \text{ nm})^2}{1239.842 \text{ eV nm}} \left(5.79 \times 10^{-5} \frac{\text{eV}}{\text{T}} \right)$$

$$\times (3.5 \text{ T}) = 0.0703 \text{ nm}$$

The wavelengths of the three normal Zeeman components are 656.112 nm, 656.112 + 0.070 nm = 656.182 nm and 656.112 - 0.070 nm = 656.042 nm.



4.

Fine structure

The energy of the 2p to 1s Lyman transition is

$$E = (-13.60570 \text{ eV}) \left(\frac{1}{2^2} - \frac{1}{1^2} \right) \\ = 10.20428 \text{ eV}$$

and the wavelength (in the absence of fine structure)

$$\lambda = \frac{hc}{E} = \frac{1239.842 \text{ eV nm}}{10.20428} = 121.5022 \text{ nm}$$

With the fine splitting of $4.5 \times 10^{-5} \text{ eV}$, the wavelength splitting is

$$\Delta \lambda = \frac{\lambda^2}{hc} \Delta E = \frac{(121.5 \text{ nm})^2}{1240 \text{ eV nm}} \times 4.5 \times 10^{-5} \text{ eV} = 0.00054 \text{ nm}$$

so

$$\lambda + \frac{1}{2} \Delta \lambda = 121.5024 \text{ nm}, \quad \lambda - \frac{1}{2} \Delta \lambda = 121.5019 \text{ nm}$$

The fine structure splits one level up by $0.5\Delta E$ and the other down by the same amount, so the wavelengths are

$$\lambda + \frac{1}{2}\Delta\lambda = 121.5024 \text{ nm}$$

and

$$\lambda - \frac{1}{2}\Delta\lambda = 121.5019 \text{ nm}$$

