

On electron power absorption in low pressure electronegative capacitive discharges

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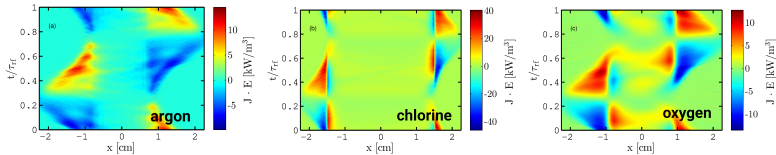
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Introduction



$p = 1.33$ Pa, $f = 13.56$ MHz, $V_0 = 222$ V, gap = 45 mm and $\gamma_{sec} = 0.0$

- The two main electron power absorption mechanisms in these discharges are
 - momentum transfer due to the moving sheaths which leads to stochastic (pressure or collisionless) heating
 - currents in the main body of the plasma discharge lead to Ohmic (or collisional) heating in the bulk and sheath regions
- We explore this using 1D particle-in-cell/Monte Carlo collision simulation for three different chemistries
 - weakly electronegative oxygen discharge – detachment
 - highly electronegative chlorine discharge – recombination
 - electropositive argon discharge

- The 1D particle-in-cell/Monte Carlo collision simulation
- The low pressure oxygen discharge
- The low pressure chlorine discharge
- Argon discharge diluted by chlorine
- Summary

The low pressure oxygen discharge



Introduction

- The oxygen chemistry is rather involved, in particular due to the presence of **metastable molecular and atomic** oxygen and their role in dissociative attachment and detachment processes
- We consider a discharge that consists of:
 - electrons
 - the ground state oxygen molecule $O_2(X^3\Sigma_g^-)$
 - the metastable oxygen molecule $O_2(a^1\Delta_g)$
 - the metastable oxygen molecule $O_2(b^1\Sigma_g)$
 - the ground state oxygen atom $O(^3P)$
 - the metastable oxygen atom $O(^1D)$
 - the negative oxygen ion O^-
 - the positive oxygen ions O^+ and O_2^+

Gudmundsson and Lieberman (2015) *Plasma Sources Sci. Technol.*, **24** 035016

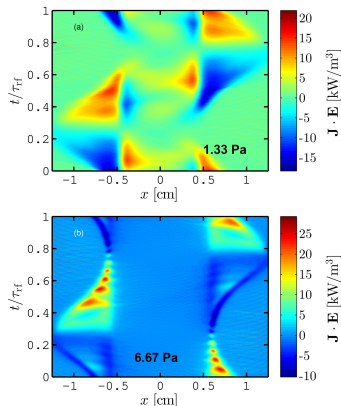
Hannesdottir and Gudmundsson (2016) *Plasma Sources Sci. Technol.*, **25** 055002

Gudmundsson and Proto (2019) *Plasma Sources Sci. Technol.*, **28** 045012



Oxygen CCP – pressure dependence

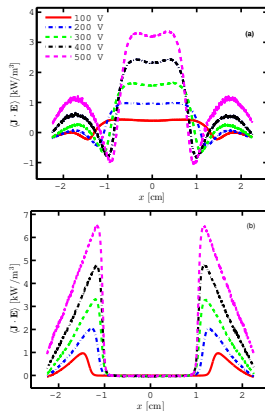
- A parallel plate capacitively coupled oxygen discharge at driving frequency of 13.56 MHz for gap separation of 45 mm
- The spatio-temporal electron heating $\mathbf{J}_e \cdot \mathbf{E}$ at 1.33 Pa (upper graph) and 6.67 Pa (lower graph)
- At 1.33 Pa there is a significant electron power absorption within the electronegative core
- At 6.67 Pa the electron power absorption occurs almost solely in the sheath region



Gudmundsson and Snorrason (2017) JAP 122 193302

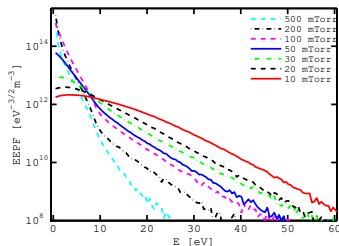
Oxygen CCP – pressure dependence

- The time averaged electron power absorption $\langle \mathbf{J}_e \cdot \mathbf{E} \rangle$ at 1.33 Pa (upper graph) and 6.67 Pa (lower graph)
- At 1.33 Pa there is significant electron power absorption within the electronegative core
- At 6.67 Pa, the electron power absorption the electronegative core is roughly zero, and the electron power absorption is almost entirely located in the sheath regions



Oxygen CCP – pressure dependence

- At low pressure (< 4 Pa) the EEPF is Druyvesteyn like and becomes bi-Maxwellian as the pressure is increased
- These results contradict what is commonly found for the capacitively coupled argon discharge where the EEPF evolves from being bi-Maxwellian at low pressure to being Druyvesteyn like at high pressure



Gudmundsson and Ventéjou (2015) JAP **118** 153302

Electron power absorption – Boltzmann term analysis

- To determine the electron power absorption mechanisms we apply Boltzmann term analysis

Surendra and Dalvie (1993) PRE **48**(5) 3914 and Schulze et al. (2018) PSST **27**(5) 055010

- The electron absorbed power can be determined as follows

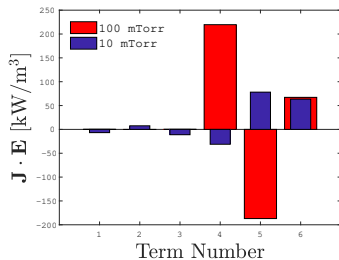
$$\begin{aligned}
 J_e \cdot E = & \underbrace{m_e u_e n_e \frac{\partial u_e}{\partial t}}_{\text{I}} - \underbrace{m_e u_e^3 \frac{\partial n_e}{\partial x}}_{\text{II}} - \underbrace{m_e u_e^2 \frac{\partial n_e}{\partial t}}_{\text{III}} \\
 & \underbrace{\hspace{10em}}_{\text{inertia}} \\
 & + \underbrace{e u_e T_e \frac{\partial n_e}{\partial x}}_{\text{IV}} + \underbrace{e n_e u_e \frac{\partial T_e}{\partial x}}_{\text{V}} + \underbrace{m_e n_e \nu_c u_e^2}_{\text{VI}} \\
 & \underbrace{\hspace{10em}}_{\text{pressure}} \hspace{10em} \underbrace{\hspace{10em}}_{\text{Ohmic}}
 \end{aligned}$$

- Terms IV and V are pressure or collisionless heating
- Term VI – electron neutral collisions or Ohmic heating



Oxygen CCP – Boltzmann term analysis

- The pressure terms are important – significant part of the electron power absorption at both 1.33 Pa and 13.3 Pa is due to the pressure terms
- The Terms IV (ambipolar) and V (electron temperature gradient) flip signs and are sharply smaller in the absolute value at 1.33 Pa
- The Ohmic term's magnitude (Term VI) is similar at both pressures
- Note that Ohmic power absorption is important even at low pressure



Proto and Gudmundsson (2020) JAP **128** 113302



The low pressure chlorine discharge



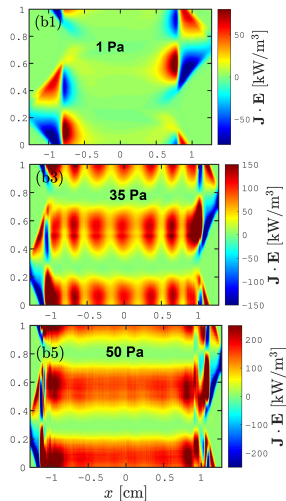
The chlorine discharge

- Chlorine is an electronegative diatomic gas that is widely used in plasma etching of both semiconductors and metals, in particular poly-silicon gate and aluminum interconnects
- We consider a discharge that consists of:
 - electrons
 - the ground state chlorine molecule $\text{Cl}_2(X^1\Sigma_g^+, v=0)$,
 - the ground state chlorine atom $\text{Cl}(3p^5^2P)$
 - the negative chlorine ion Cl^-
 - the positive chlorine ions Cl^+ and Cl_2^+



Electron power absorption

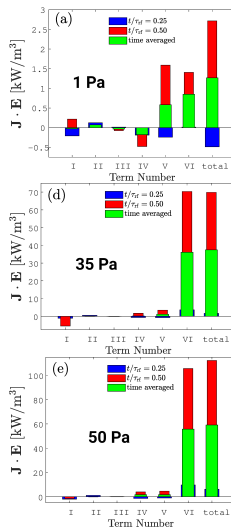
- The spatio temporal behavior of the total electron power absorption $\mathbf{J}_e \cdot \mathbf{E}$ over the full gap length for a capacitively coupled chlorine discharge
- At 1 Pa there is clear sign of drift ambipolar heating (DA-mode) and stochastic heating (α -mode)
- At 35 and 50 Pa there are indications of striations in addition to DA- and α -mode)
- DA power absorption is expected within the electronegative core



Electron power absorption

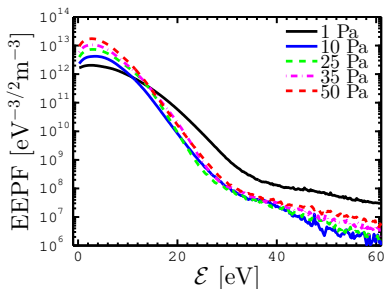
- The space averaged electron power absorption profile terms
 - $t/\tau_{rf} = 0.25$ blue bar
 - $t/\tau_{rf} = 0.50$ red bar
 - time averaged green bar
- At 1 Pa the pressure terms and the Ohmic term contribute to the electron power absorption
- At higher pressures Ohmic power absorption dominates

Proto and Gudmundsson (2021) PSST **30**(6) 065009



Electron power absorption

- The electron energy probability function (EEPF) in the discharge center is Druyvesteyn like at all pressures
- This is expected when there is significant Ohmic heating in the plasma bulk

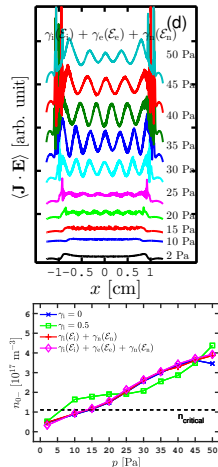


Striations

- Striations are known to appear in electronegative discharges when two conditions are simultaneously fulfilled:
 - high enough electronegativity
 - driving frequency that is comparable to the ion plasma frequency
- Based on the ion-ion plasma model striations may appear when $n_{\text{ion}} > n_{\text{critical}}$ where the critical density is defined as

$$n_{\text{critical}} = \frac{\omega_{\text{rf}}^2 \epsilon_0 \mu}{e^2}$$

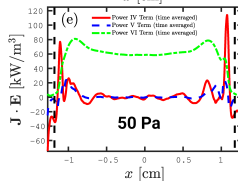
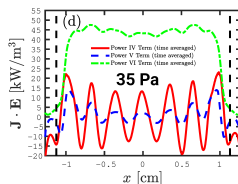
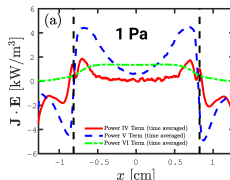
where $\mu = m_+ m_- / (m_+ + m_-)$ is the reduced mass of the positive and negative ions



Mahdavi pour and Gudmundsson (2025)

Striations - electron power absorption

- The time averaged electron power absorption profile of
 - term IV (red line)
 - term V (blue dashed line)
 - term VI (green dot dashed line)
- At 1 Pa the pressure terms and the Ohmic term contribute to the electron power absorption
- At higher pressures Ohmic power absorption dominates
- At 35 Pa striations are observed – the pressure terms are apparent



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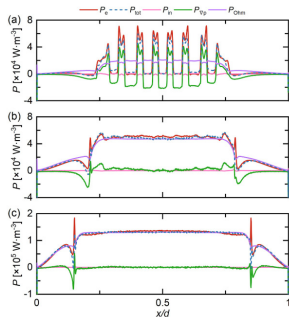
Striations – electron power absorption

- Similar findings have been reported more recently for a capacitive CF₄ discharge operated at 60 Pa and 6.78 MHz with 2 cm electrode gap (top graph)
- In this case the pressure terms

$$P_{\nabla p} = \underbrace{eu_e T_e \frac{\partial n_e}{\partial x}}_{\text{IV}} + \underbrace{en_e u_e \frac{\partial T_e}{\partial x}}_{\text{V}}$$

dominate the power absorption process

- For higher operating frequency the Ohmic term dominates the electron power absorption (lower graphs)



Zhou et al. (2025) PSST **34**(8) 085014

Argon discharge diluted by chlorine

Current source operated at a single frequency

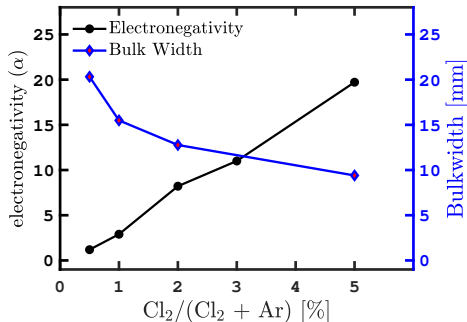
$$J(t) = J_{\text{rf}} \sin(2\pi ft)$$

gap = 2.54 cm, $J_{\text{rf}} = 50 \text{ A/m}^2$ and $f = 13.56 \text{ MHz}$, $p = 6.67 \text{ Pa}$



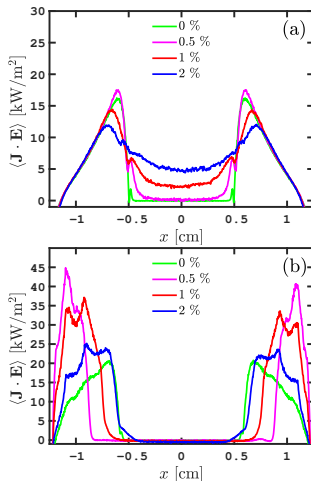
Argon discharge diluted by chlorine

- Ar/Cl₂ mixture at 6.67 Pa
- The electronegativity increases with increased chlorine fraction
- The plasma bulk width decreases with increased chlorine fraction
- Gap width 2.5 cm, driving current density 50 A/m² and frequency 13.56 MHz
- The argon model includes excited argon atoms



Argon discharge diluted by chlorine

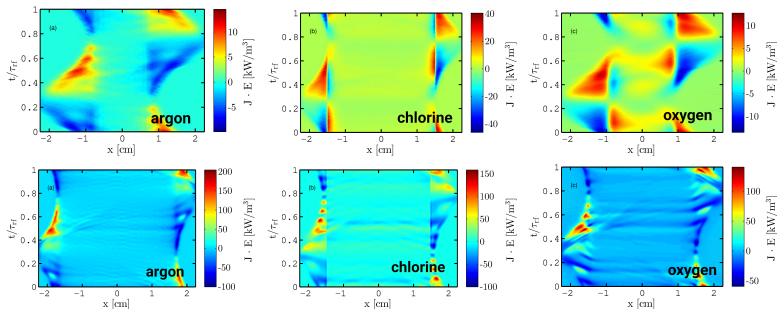
- Ar/Cl₂ mixture at 6.67 Pa
- The time-averaged electron power absorption for varying chlorine dilution
 - (a) neglecting excited argon states
 - (b) including excited argon states



Summary

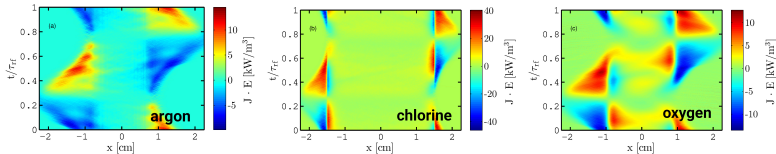


Summary



- Capacitively coupled argon, oxygen and chlorine discharges were studied by particle-in-cell/Monte Carlo collision simulations

Summary



- In a chlorine discharge DA electron power absorption dominates and becomes increasingly more Ohmic with increased pressure
- Including the detachment processes by the singlet metastable states has a strong influence on the electronegativity in the oxygen discharge
- A heating mode transition, from hybrid drift-ambipolar (DA) and α -mode to pure α -mode, is observed as the pressure, driving frequency, or electrode gap spacing are increased, or the quenching coefficient of the oxygen metastable $O_2(a^1\Delta_g)$ is varied



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Thank you for your attention

The slides can be downloaded at

<http://langmuir.raunvis.hi.is/~tumi/ranns.html>

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