Electron power absorption in low pressure capacitively coupled discharges of complex chemistry

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- The capacitively coupled discharge is sustained by applying radio-frequency (rf) voltage or current between electrodes and a neutral gas at low pressure is injected between the electrodes
- A high-voltage sheath forms between the plasma bulk and the electrodes
- The two main electron power absorption mechanisms in these discharges are
 - Ohmic (or collisional) heating in the bulk and sheath regions
 - momentum transfer due to the moving sheaths which leads to stochastic (pressure or collisionless) heating
- We explore this for three different chemistries
 - electropositive argon discharge
 - weakly electronegative oxygen discharge
 - highly electronegative chlorine discharge.





The electron power absorption

- *p* = 10 mTorr
- f = 13.56 MHz
- gap = 45 mm
- $\gamma_{\text{see}} = 0.0$
- For argon the discharge is operated in α-mode
- Chlorine and oxygen discharges operate in hybrid α- and drift ambipolar (DA)-mode







The electron power absorption

- *p* = 10 mTorr
- *f* = 27.12 MHz
- gap = 45 mm
- $\gamma_{\text{see}} = 0.0$
- The argon and oxygen discharges are operated in α-mode
- The chlorine discharge operated in hybrid α- and drift ambipolar (DA)-mode





The oxygen discharge



- Oxygen forms a weakly electronegative discharge
- The oxygen chemistry is rather involved, in particular due to the presence of metastable molecular and atomic oxygen and their role in dissociative attachment and detachment processes
- We use the 1d3v oopd1 (objective oriented plasma device for one dimension) particle-in-cell Monte Carlo collision code to simulate the discharge

Gudmundsson et al., Plasma Sources Sci. Technol., 22 035011 (2013)



The oxygen discharge

We consider a discharge that consists of:

- electrons
- the ground state oxygen molecule O₂(X³Σ_g)
- the metastable oxygen molecule O₂(a¹∆_g)
- the metastable oxygen molecule O₂(b¹Σ_g)
- the ground state oxygen atom O(³P)
- the metastable oxygen atom O(¹D)
- the negative oxygen ion O⁻
- the positive oxygen ions O⁺ and O⁺₂
- The discharge model includes energy dependent secondary electron emission yield
- We apply a global model¹ beforehand to calculate the partial pressure of the various neutrals

Thorsteinsson and Gudmundsson, Plasma Sources Sci. Technol., 19 055008 (2010)



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- The negative ion density decreases as the metastables O₂(a¹Δ_g) and O₂(b¹Σ_g) are added to the discharge model
- The electron power absorption in the plasma bulk drops to zero at the higher pressures
- The effective electron temperature profile changes significantly when detachment by singlet metastables is added to the reaction set

10 mTorr, 50 mTorr and 200 mTorr

Gudmundsson and Hannesdottir, AIP Conf. Proc. 1811 120001 (2017)

Gudmundsson and Lieberman (2015) PSST 24 035016



- A parallel plate capacitively coupled oxygen discharge at driving frequency of 13.56 MHz for gap separation of 45 mm
- The spatio-temporal electron heating J_e · E at 10 and 50 mTorr
- At 10 mTorr there is a significant electron heating within the electronegative core
- At 50 mTorr the electron heating occurs almost solely in the sheath region

Hannesdottir and Gudmundsson (2016) PSST, 25 055002

Gudmundsson and Ventéjou (2015) JAP 118 153302



- The time averaged electron heating (J_e · E) at 10 and 50 mTorr
- At 10 mTorr there is significant electron heating within the electronegative core
- At 50 mTorr, the heating rate in the electronegative core is roughly zero, and electron heating is almost entirely located in the sheath regions

Gudmundsson and Snorrason (2017) JAP 122 193302



- The axial electric field at *t*/*τ*_{rf} = 0.5 for both 10 and 50 mTorr
- At 10 mTorr there is a significant electric field strength within the electronegative core while at 50 mTorr it is zero
- This strong electric field within the plasma bulk (the electronegative core) indicates a drift-ambipolar (DA) heating mode
- The electronegativity is significantly higher when operating at 10 mTorr α > 20 than when operating at 50 mTorr α < 0.1





Oxygen CCP – frequency dependence



- At 12 MHz significant heating is observed in the plasma bulk but also in the sheath region
- At 19 MHz the heating and cooling in the sheath regions has increased, however there is contribution to the electron heating in the bulk region (note the change in scale)

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 At 20 MHz there is almost no electron heating in the plasma bulk

Oxygen CCP – electrode spacing

- A parallel plate capacitively coupled oxygen discharge at 50 mTorr driven by a 400 V voltage source at 13.56 MHz as the gap separation is varied
- At small electrode spacing a combination of stochastic (α-mode) and drift ambipolar (DA) heating in the bulk plasma (the electronegative core) is observed
- The DA-mode dominates the time averaged electron heating
- As the electrode spacing is increased, the heating mode transitions into a pure α-mode



Oxygen CCP – electrode spacing

- The time averaged electron power absorption (J_e · E)
- The electron density in the discharge center (left y axis) and the electronegativity in the discharge center (right y axis) as a function of the gap separation
- The heating mode transition coincides with a sharp decrease in electronegativity
- This agrees with recent experimental findings of You et al. (2019)



Oxygen CCP – Bolzmann term analysis

 To explore the electron power absorption processes we apply Boltzmann term analysis
Schulze et al. (2018) PSST 27 055010

$$E = -\underbrace{\frac{m_{e}}{e}\frac{\partial u_{e}}{\partial t}}_{I} + \underbrace{\frac{m_{e}}{e}\frac{u_{e}^{2}}{n_{e}}\frac{\partial n_{e}}{\partial x}}_{II} + \underbrace{\frac{m_{e}}{e}\frac{u_{e}}{n_{e}}\frac{\partial n_{e}}{\partial t}}_{III} - \underbrace{\frac{T_{e}}{n_{e}}\frac{\partial n_{e}}{\partial x}}_{IV} - \underbrace{\frac{\partial T_{e}}{\partial x}}_{V} - \underbrace{\frac{m_{e}u_{e}\nu_{c}}{e}}_{VI}$$

- The pressure terms are important
- The Terms IV (ambipoar) and V (electron temperature gradient) flip signs and are sharply smaller in the absolute value at 10 mTorr
- The Ohmic term's magnitude (Term VI) is similar at both pressure



The chlorine discharge



Chlorine CCP

- Chlorine is an electronegative diatomic gas that is widely used in plasma etching of both semiconductors and metals, in particular poly-silicon gate and aluminum interconnects
- Chlorine atoms are believed to be the primary reactant in plasma etching
 - The chlorine molecule has a low dissociation energy (2.5) eV)
 - a near-zero threshold energy for dissociative attachment
- We consider a discharge that consists of:
 - electrons
 - the ground state chlorine molecule $Cl_2(X^{1}\Sigma_{\sigma}^{+}, v = 0)$,
 - the ground state chlorine atom Cl(3p^{5 2}P)
 - the negative chlorine ion Cl⁻
 - the positive chlorine ions Cl⁺ and Cl⁺₂



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Chlorine CCP



- There is significant electron power absorption (red and yellow areas) and smaller power loss (dark blue areas) evident within the plasma bulk
- This indicates that the electron power absorption occurs predominantly through DA electron power absorption



Chlorine CCP



- The Ohmic term has the largest contribution to electron power absorption at 1Pa
- All the terms contribute to the electron power absorption
- The time averaged terms IV (ambipolar) and V (electron temperature gradient) have opposite signs



Summary



Summary



 Capcacitively coupled argon, oxygen and chlorine discharges were studied by particle-in-cell/Monte Carlo collision simulations

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Summary



- In a chlorine discharge DA electron power absorption dominates and becomes increasingly more Ohmic with increased pressure
- Including the detachment processes by the singlet metastable states has a strong influence on the electronegativity in the oxygen discharge
- A heating mode transition, from hybrid drift-ambipolar (DA) and α-mode to pure α-mode, is observed as the pressure, driving frequency, or electrode gap spacing are increased, or the quenching coefficient of the oxygen metastable O₂(a¹Δ_g) is varied

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Thank you for your attention

The slides can be downloaded at http://langmuir.raunvis.hi.is/~tumi/ranns.html

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