Singlet Oxygen

production, photodissociation, and detection

in discharges, clusters and ice surfaces using velocity map imaging



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Molecular Oxygen

- Photodissociation (W. vd Zande, X.M. Yang,...)
- REMPI (C. Western)
- O₂-X Clusters (A. Baklanov) and CIA
- Inelastic Scattering with H₂, He, Ar (F. Lique)
- Photodesorption of O₂-ice at 10K (H. Linnartz)

Electronic structure and spectrum of O₂





Velocity map imaging of ions and electrons using electrostatic lenses: Application in photoelectron and photofragment ion imaging of molecular oxygen

Authors André TJB Eppink, David H Parker

Publication date	1997/9/1
Journal	Review of Scientific Instruments
Volume	68
Issue	9
Pages	3477-3484
Publisher	AIP Publishing
Description	The application of electrostatic lens

Scription The application of electrostatic lenses is demonstrated to give a substantial improvement of the two-dimensional (2D) ion/electron imaging technique. This combination of ion lens optics and 2D detection makes "velocity map imaging" possible, ie, all particles with the same initial velocity vector are mapped onto the same point on the detector. Whereas the more common application of grid electrodes leads to transmission reduction, severe trajectory deflections and blurring due to the non-point source geometry, these problems ...



$AB^* \longrightarrow A + B$



parallel transition (sin² θ) β = 2



Electronic structure and spectrum of O₂



Detection of cold O_2 X-state by the O(³P₂) image



O(³P₂) (2+1) REMPI at 225.6 nm

Photodissociation of cold X-state O₂ via the HC





Deconvolution of the Herzberg Continuum



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Detection of O_2 b-state by the O(³P₂) image



good test of aparatus resolution (\imited by electron recoil)

Detection of O_2 b-state by the O(³P₂) image



Fine Structure Ratios: $O({}^{3}P_{J})$ J=2:1:0 are 1.00 (+/-0.05) : 0 : 0

At threshold β = 2.0, no evidence of alignment effects

Detection of O_2 X-state by the $O({}^1D_2)$ image



Detection of O_2 b-state by the $O({}^1D_2)$ image



Extra rings are due to (2+1) rempi of b state, dissociation of O_2^+

Maximal M₁=0 production is seen via the b-state

Cluster photodissociation $O_2 - X + UV \rightarrow O_2 (X, a, b) + X$ $O_2^- + X^+$ $O_2 - X^+$

Detection: Absorption, Emission, REMPI ? We use: 2+1 REMPI of O(³P_{2,1,0}) at 226 nm

THE O(³P) IMAGE IS A SIGNATURE OF MANY POSSIBLE FORMS OF O₂

Formation of O₂ $a(^{1}\Delta)$ from O₂-X 1:1 van der Waals clusters

 $O_2 + hv \rightarrow O + O$ T_{kin}= 0.183 eV \vec{E} 02 Photodissociation of van der Waals clusters of isoprene with oxygen, $C_5H_8-O_2$, in the wavelength range 213–277 nm Konstantin V. Vidma, Pim W. J. M. Frederix, David H. Parker, and Alexey V. Baklanov J. Chem. Phys. 137, 054305 (2012)

>330 enhancement in the O(³P₂) yield !!!



 $O({}^{3}P_{2})$ from photodissociation of $O_{2} b^{1}\Sigma_{\sigma}$

Production of maximally aligned $O(^{1}D)$ atoms from two-step photodissociation of molecular oxygen

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Symmetry breaking

⊥/∥	forbidden by:
mixed	parity ($\sigma_{mol \ plane}$)
	spin
	Δ L=2 (press. dep.)
	⊥ / mixed ⊥ ⊥



The Herzberg III band is enhanced by collisions (x50 at 1 bar)

Ice surfaces and astrochemistry



Studying ice surface photochemistry



Procedure

- 1. Grow Ice surface
- 2. Characterize by
 - a) RAIRS
 - b) TPD-QMS
- 3. Irradiate (pulsed or cw)
- 4. Characterize by
 - a) RAIRS
 - b) TPD-QMS
 - c) post-ioniz. TOF-MS
 - d) VMI





Desorption of O(³P₂) from 15K O₂-ice at 250 nm



O(¹D) signatures 250 nm desorption



REMPI spectrum in a-state region



REMPI spectrum: confirmation of O-a(${}^{1}\Delta_{g}$) state



Desorption of O(³P₂) from 20K and 140K O₂-ice at 320 nm



Finally, O₂ discharges

A study of arrested development



PCCP

PAPER

Photodissociation of singlet oxygen in the UV region

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Zahid Farooq, Dimitri A. Chestakov, Bin Yan, Gerrit C. Groenenboom, Wim J. van der Zande and David H. Parker* 4.0 mm

M.van Beek & J.J ter Meulen CPL 2001



OH flux = $2 \cdot 10^{17}$ #/ster.sec !!

Experiment: Pulsed expansion of water/Ar, HV discharge pulse during middle of expansion, skimmed, cross with one or two UV lasers, image $O({}^{3}P_{J})$ products. Turn off channelplates during discharge to avoid looking at XUV photons. Lots of cold $O({}^{3}P_{2})$ atoms formed, and Fe* from electrodes.







Photodissociation of singlet oxygen in the UV region

Zahid Farooq, Dimitri A. Chestakov, Bin Yan, Gerrit C. Groenenboom, Wim J. van der Zande and David H. Parker* *Phys. Chem. Chem. Phys.*, 2014, **16**, 3305



Energy distributions $O({}^{3}P_{i})$ from O_{2} discharge



Photodissociation cross sections for the O₂ X, a, b states



Conclusions

- Many forms of O₂ can be identified in an O atom image
- Enhanced absorption in O₂-van der Waals clusters efficiently produces singlet oxygen in a spin flip process
- Laser desorption at 10K O₂-ice surfaces leads to atom recombination to produce significant amounts of singlet oxygen, along with O atoms and hot X O₂
- Atom recombination in discharge produces a significant amount of singlet oxygen, hot ground state molecules, and O(³P₂) atoms









Thanks for your attention!

There are many more non-bound states than bound states!



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$O_2 + h\nu \rightarrow O(^{3}P) + O(^{3}P)$

Fine structure dissociation limits



(2+1) REMPI detection scheme

