

Quantitative Diagnostics of Inductively-Coupled Plasmas in O_2 : Densities and energy distributions



Laboratoire de Physique des Plasmas

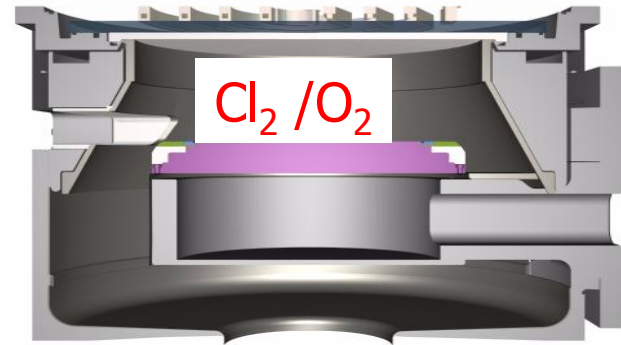
Jean-Paul Booth,
Mickäel Foucher, Daniil Marinov, Andrew Gibson,
Adriana Annušová and Vasco Guerra



Motivation



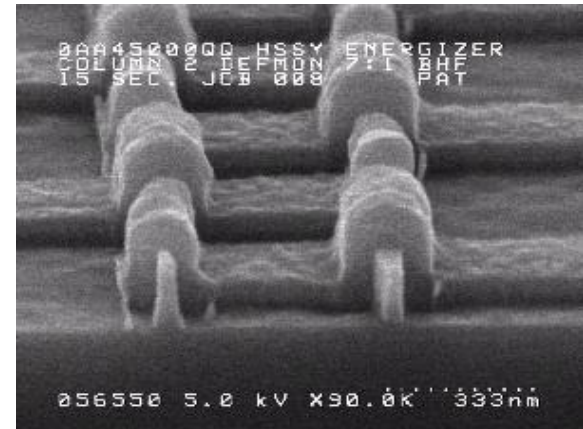
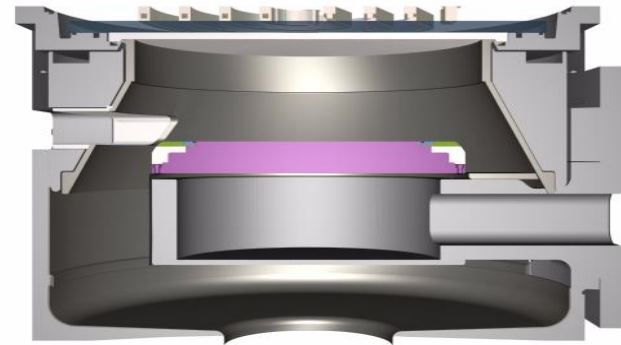
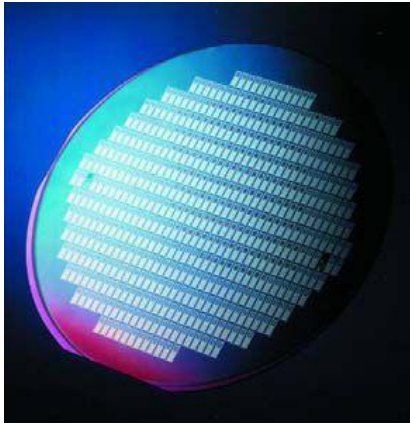
Why study low pressure RF inductive discharges in diatomic gases?



Motivation



RF plasmas in Cl_2 / O_2 (& often HBr) at low pressure widely used for selective, anisotropic etching of Si, InP etc

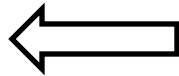


Motivation: understand plasmas in molecular gases



Most academic studies..

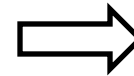
Rare gas
plasmas:
Ar



Most applications..

Mixtures of polyatomic gases:

- CF_4
- $\text{C}_4\text{F}_8/\text{O}_2/\text{Ar}$
- SiH_4/H_2
- $\text{Cl}_2/\text{HBr}/\text{O}_2$

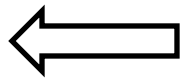


Motivation: understand plasmas in molecular gases

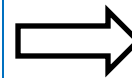


Most plasma physics studies..

Rare gas
plasmas:
Ar



Diatomic gas
plasmas:
O₂, Cl₂, H₂...



Most applications..

Mixtures of polyatomic gases:
•CF₄
•C₄F₈/O₂/Ar
•SiH₄/H₂
•Cl₂/HBr/O₂

Shows most of the mechanisms occurring in polyatomics:

- Dissociation, surface recombination
- Electronegativity
- Vibrational + rotational excitation

-But simpler, can measure (nearly) everything!

Test (validate) and improve Models:

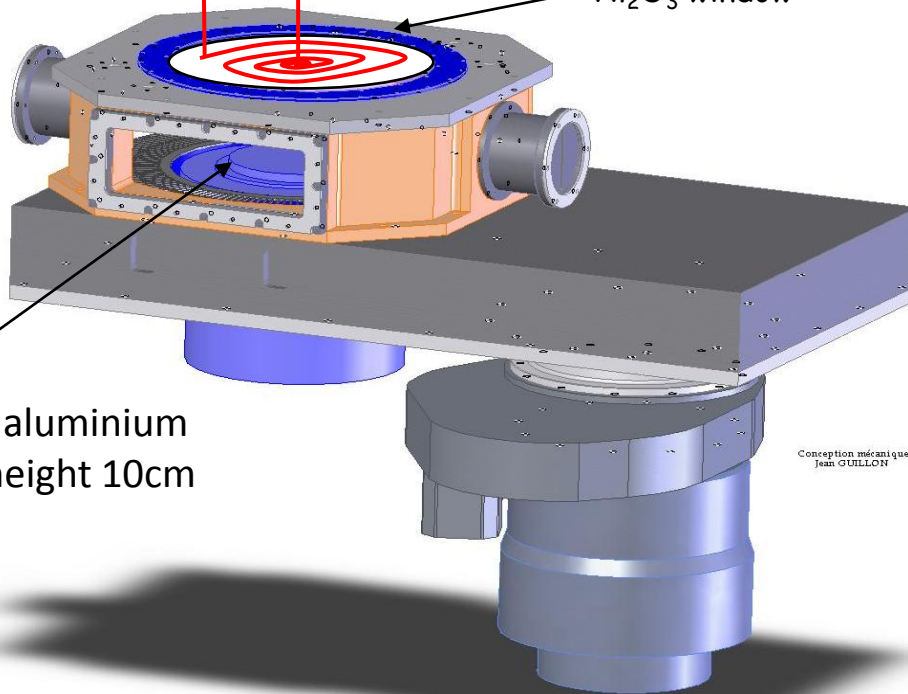
- “Global” 0D models with plasma chemistry
- **2D Fluid Plasma + chemistry model : HPEM**

The Inductively Coupled Plasma Reactor at LPP:



13.56 MHz
power supply

4 turn spiral antenna
Al₂O₃ window



Chamber:
Anodized aluminium
Ø 55cm height 10cm

Conception mécanique:
Jean CILLICON

- Industrial Scale Reactor
dimension for 300mm wafers
- Industrial gases (O₂, Cl₂, HBr)

Pressure : 5-100 mTorr
Power : up to 500W

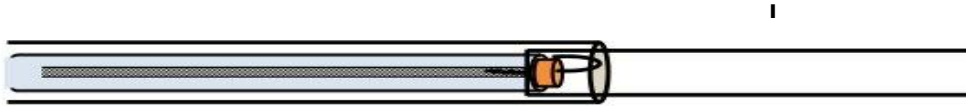
- All surfaces Al₂O₃
(no substrate)

What we can measure?



Electron density	-	Microwave resonant probe
Negative ions	-	+ laser photodetachment
O atom density	-	TALIF
Vibrational distribution-		UV absorption spectroscopy
Gas temperature:		
O ₂ Trot	-	UV absorption spectroscopy
Ar ^m Doppler	-	IRLAS vs IRLIF
O atom Doppler-		HR TALIF

Electron density : Hairpin probe



$\frac{1}{4}$ wave resonator : ~ 3 GHz :

- measure plasma permittivity from frequency shift with plasma
- deduce electron density from permittivity

Avoids many of the problems of Langmuir probes:

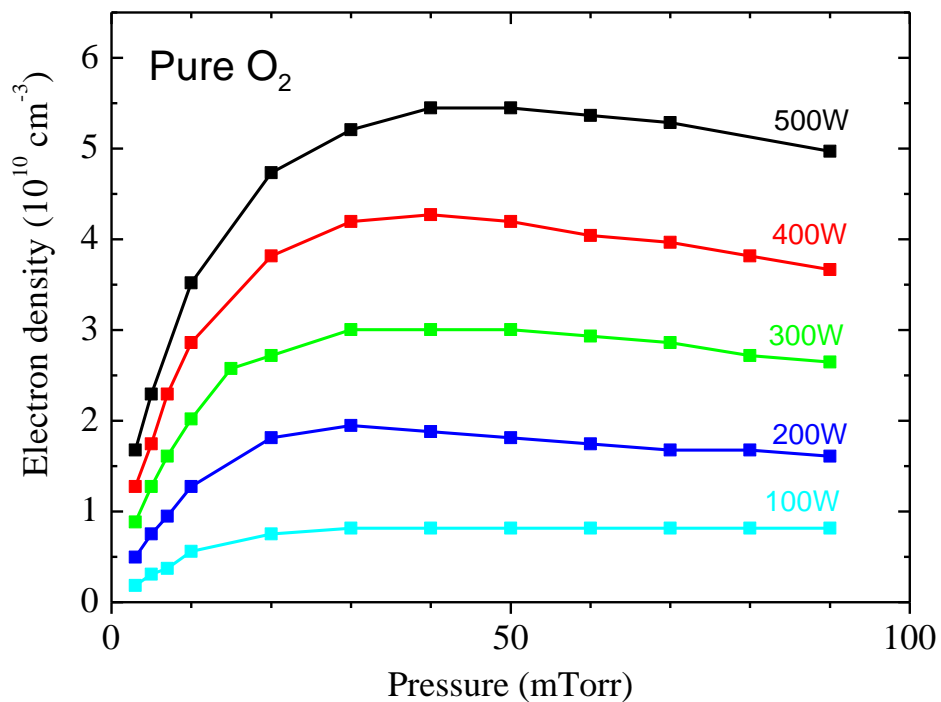
- Probe contamination
- Return current path (insulating reactor walls!)
- RF compensation

Negative ions : detect electron pulse from laser
photodetachment

Electron density



Broad maximum @ 40 mTorr

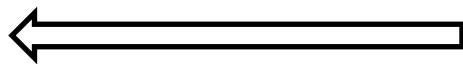
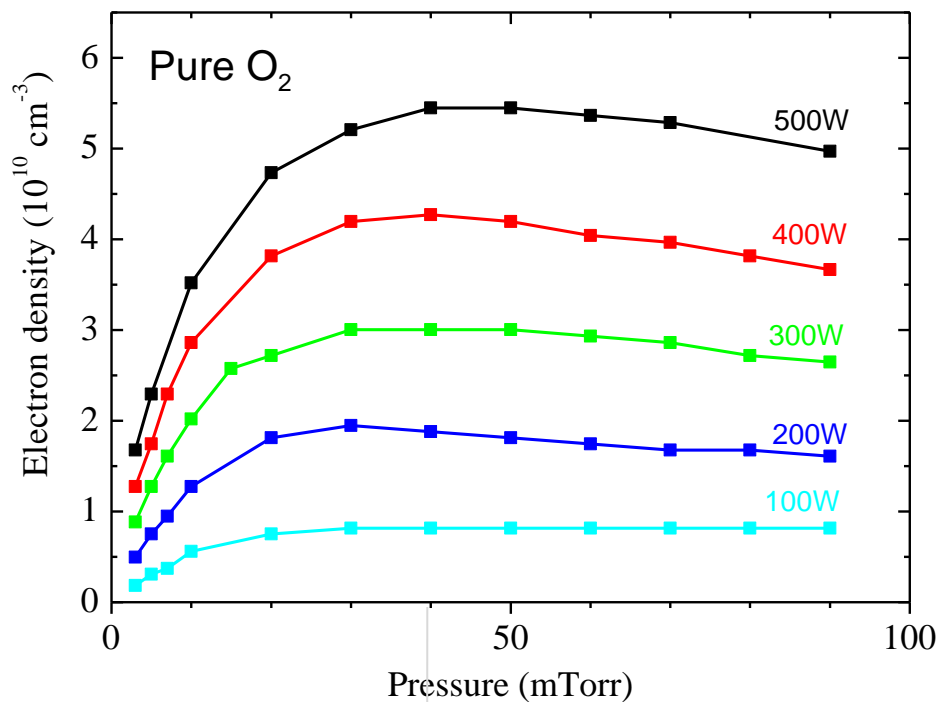


Approximately linear increase with RF power

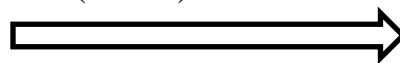
Electron density



Broad maximum @ 40 mTorr



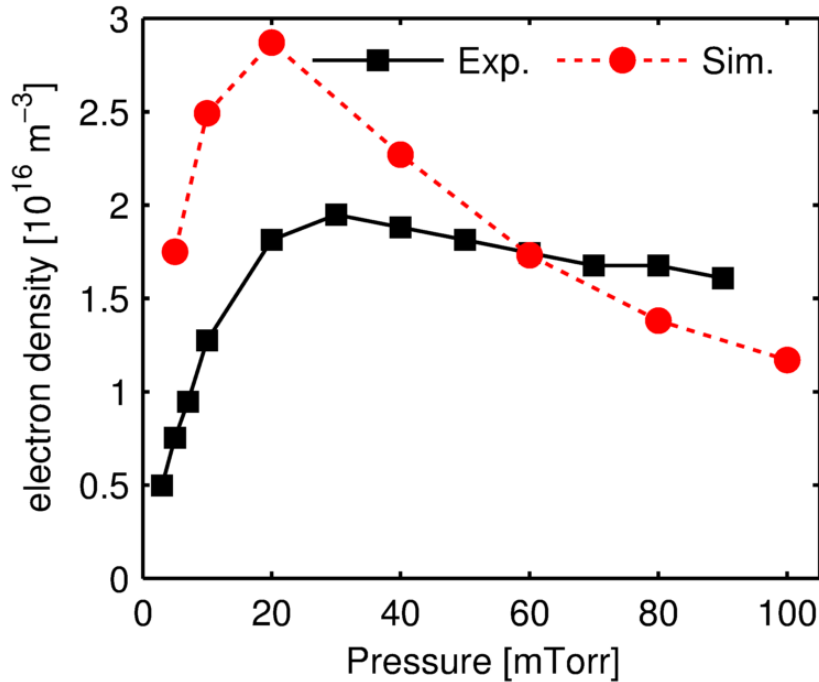
High Te, Vp:
Energy dissipated
accelerating ions
across sheath



Energy dissipated in inelastic collisions

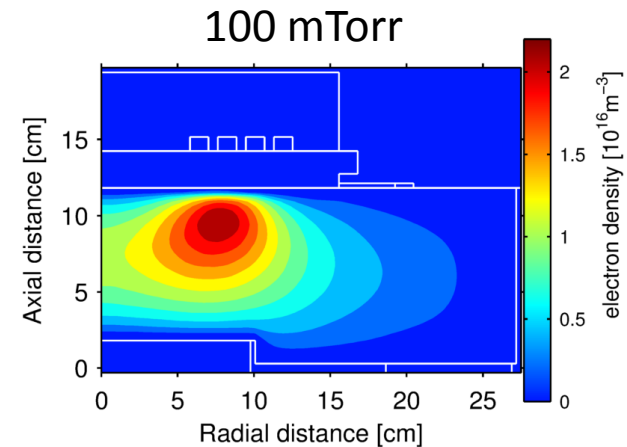
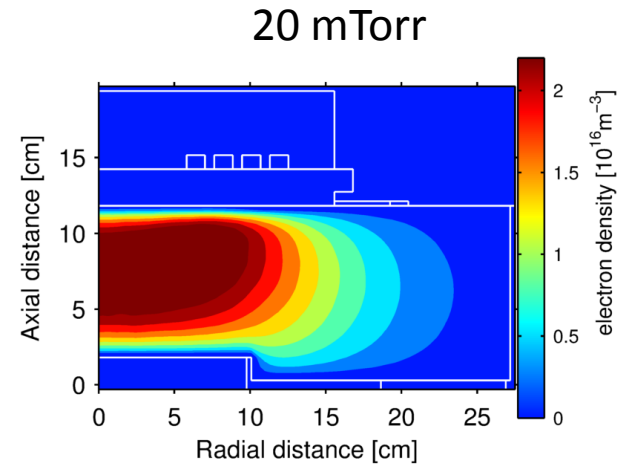
Negative ions ?

Electron density :O₂ 200 W Compare to HPEM



Predicts maximum at the correct pressure

High pressure trend poorly modelled

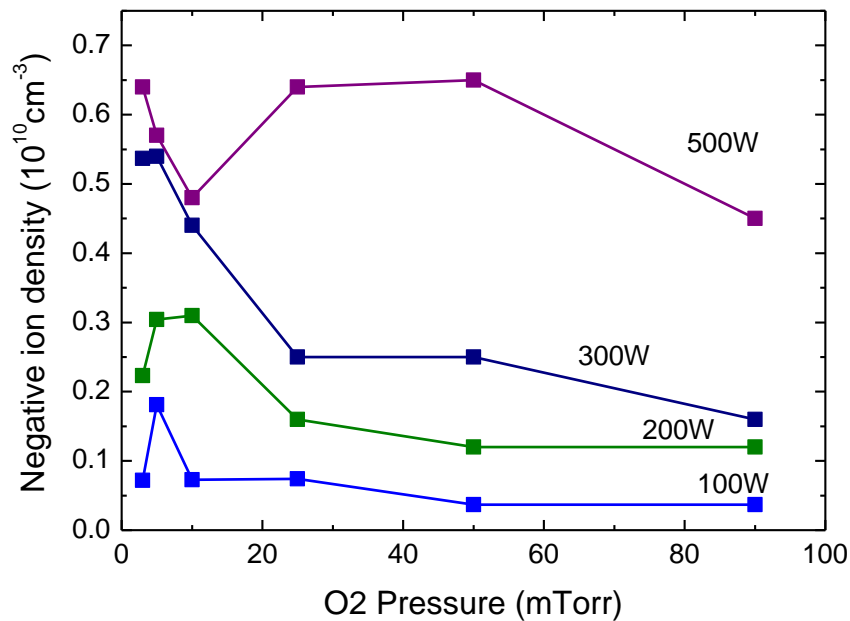


Electron density moves off-axis

Negative Ion density



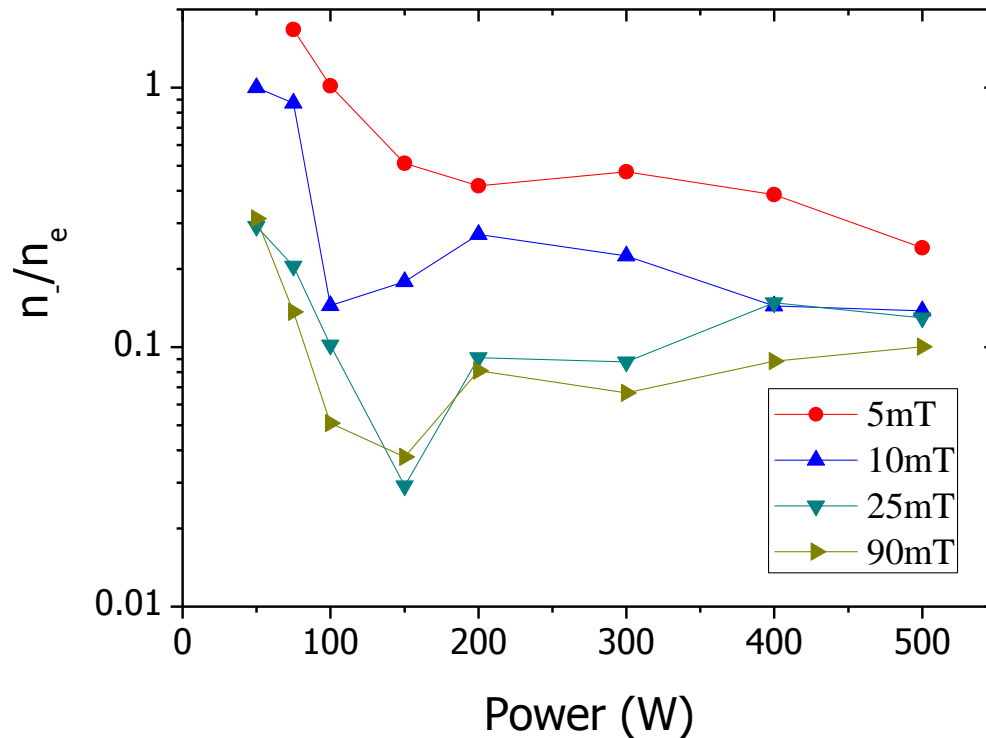
Laser photodetachment /
hairpin detection of photoelectrons



Courtesy of Nishant Sirse,
Dublin City University

Electronegativity

$$\alpha = n_- / n_e$$



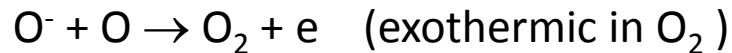
Negative ion density lower than electron density except at **low pressure and power**

Not responsible for electron depletion

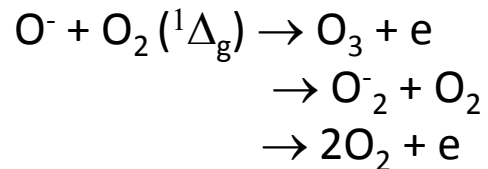
Negative ion destruction : Associative Detachment



Atoms:



Metastable states : $\text{O}_2 \ ^1\Delta_g$ (at ≈ 1 eV)

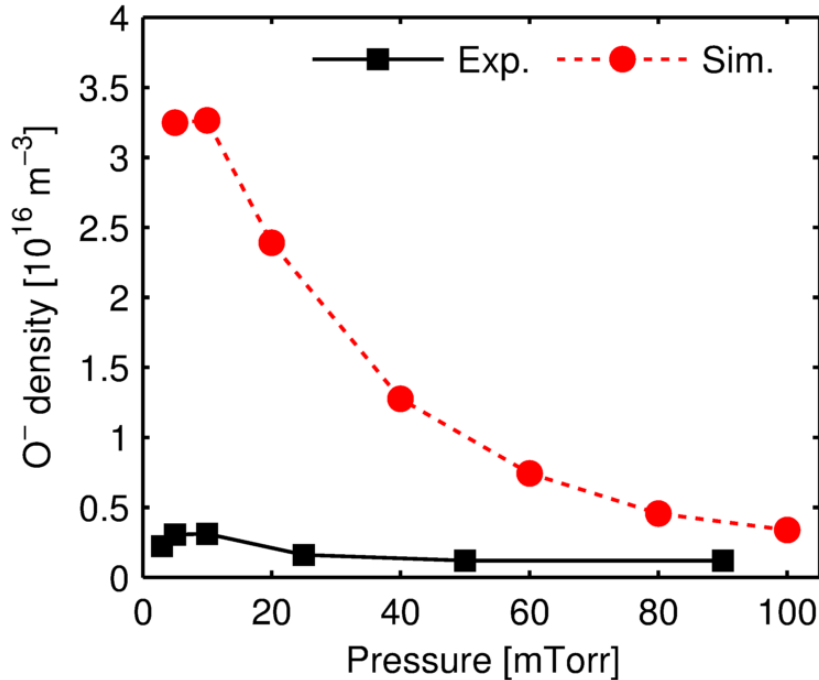


Midey et al J. Phys. Chem. A 2008, 112, 3040-3045

As a result, O_2 plasmas have a much lower density of negative ions

Electro-negativity highest at low pressure and low power (O and $\text{O}_2 \ ^1\Delta_g$ low)

Negative ion density: compare to HPEM Simulation



Pure O_2

Model strongly overestimates
negative ion density!

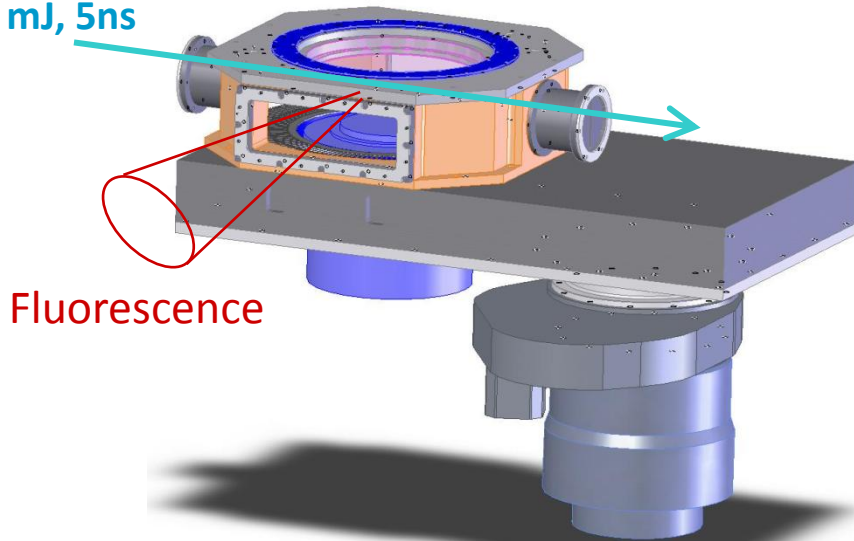
Loss processes underestimated, or
missing mechanisms?

Atom densities:

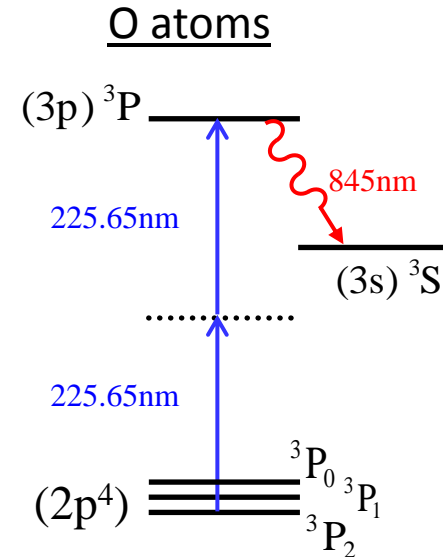
Two-Photon Absorption Laser-Induced Fluorescence (TALIF)



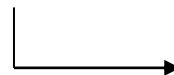
Pulsed
UV laser
1 mJ, 5ns



Fluorescence

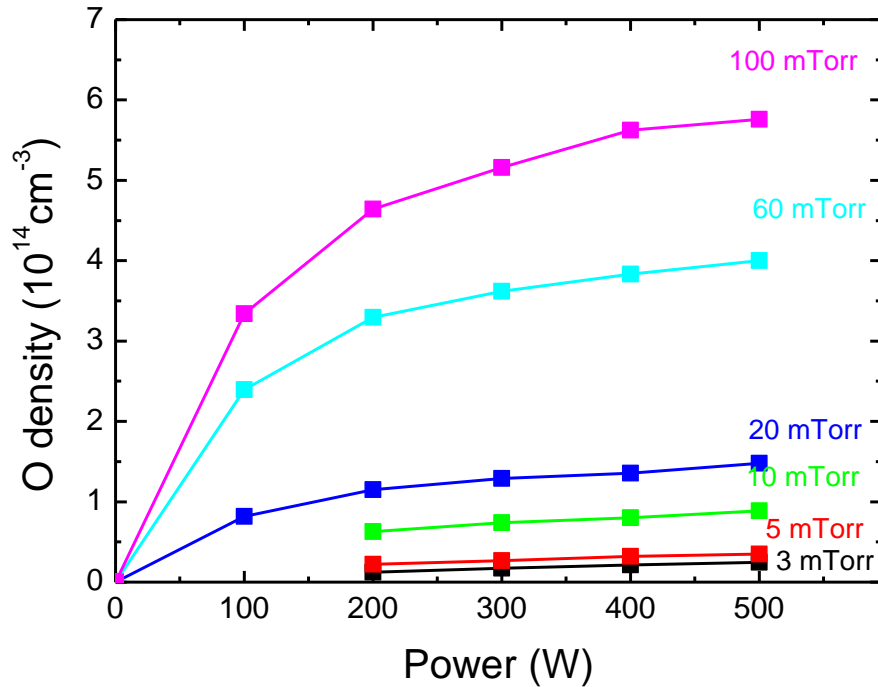


- High spatial and temporal resolution measurements
- Relative densities of ground-state O atoms
- Absolute densities: use calibration techniques



Niemi et al : PSST **14**(2005) 375-386

Pure O₂ : Atom density



O density increases:

-with pressure

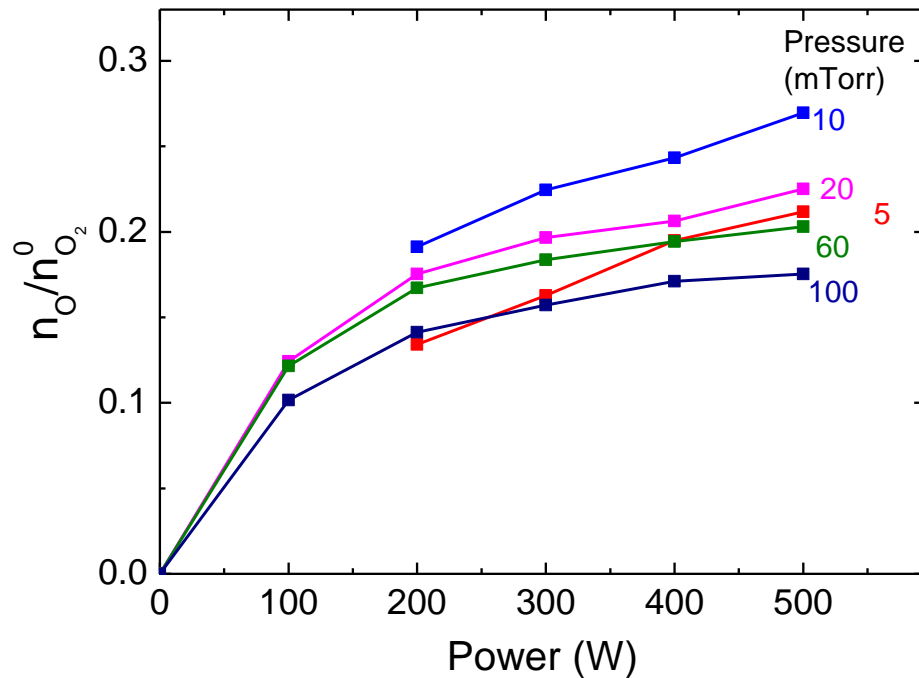
-with RF power :but saturates

→Dissociation fraction?

Pure O₂ : "Dissociation fraction "



Normalise to $n_{O_2}^0$, the density of (cold) gas before plasma:



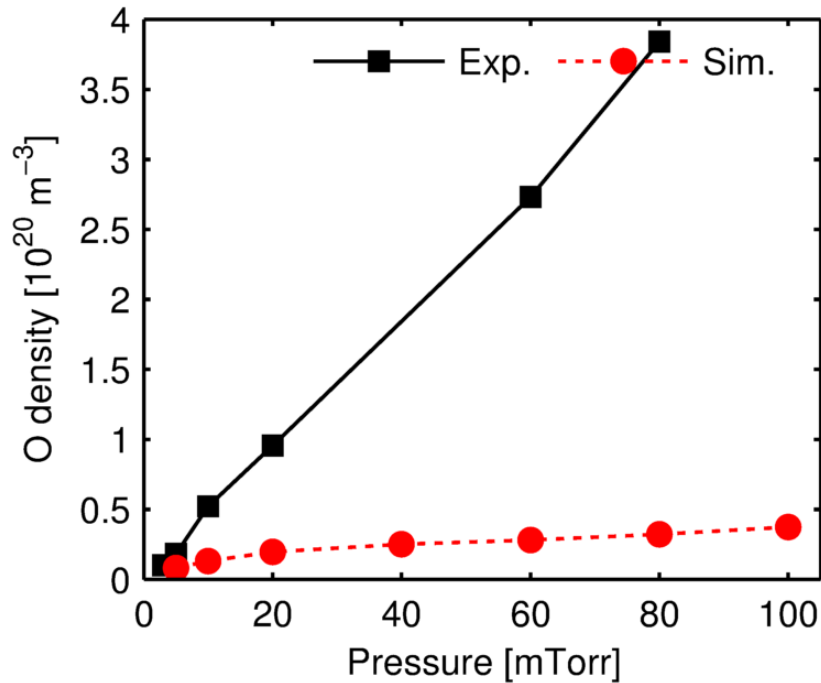
Maximal at 10 mTorr
(n_e maximum at 40 mTorr)

Dissociation saturates @ 20-30%

Why not 100% ?

Gas temperature?

O atom density: comparison to HPEM

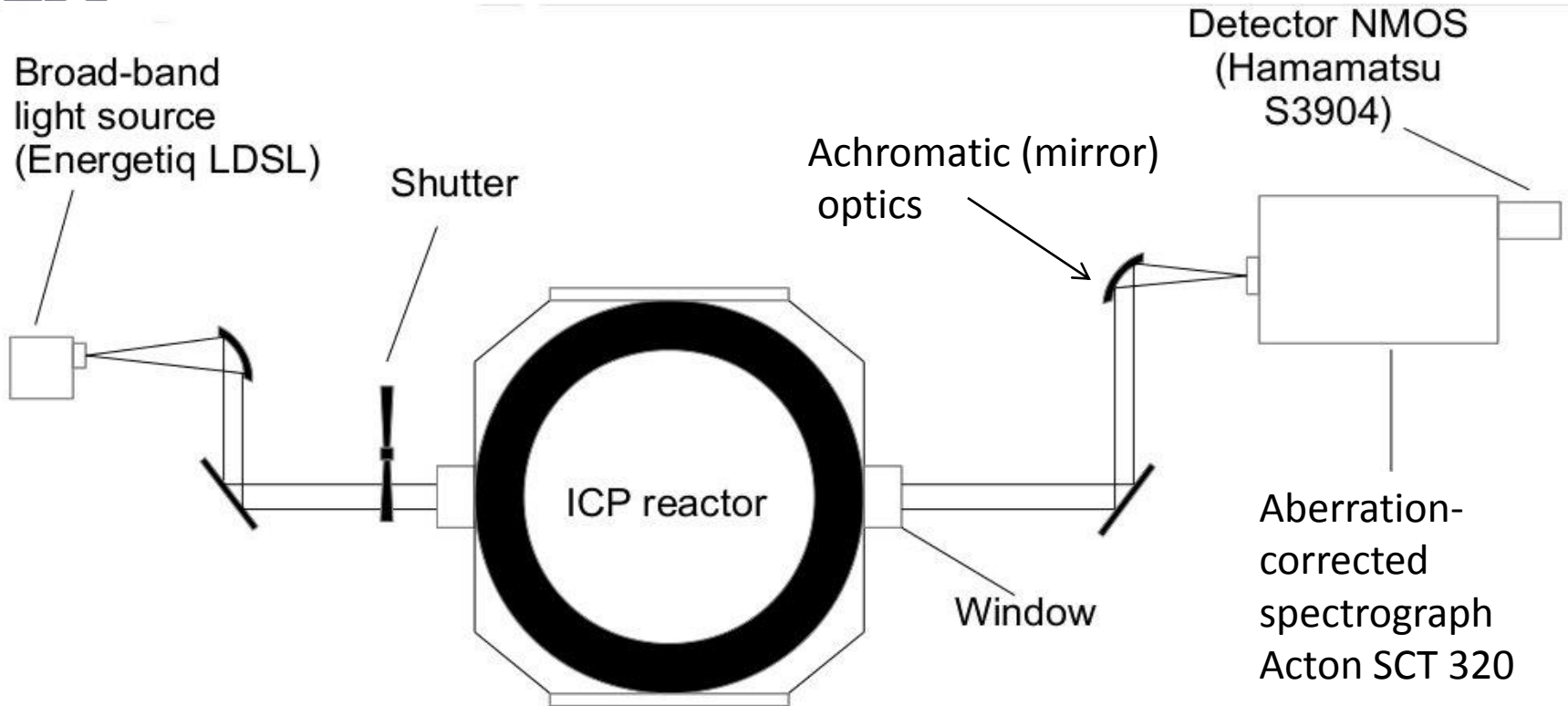


Pure O_2 200 W

Model strongly overestimates O
atom density!

Error in dissociation cross-
sections?

High-sensitivity ultra-broad-band Absorption spectroscopy



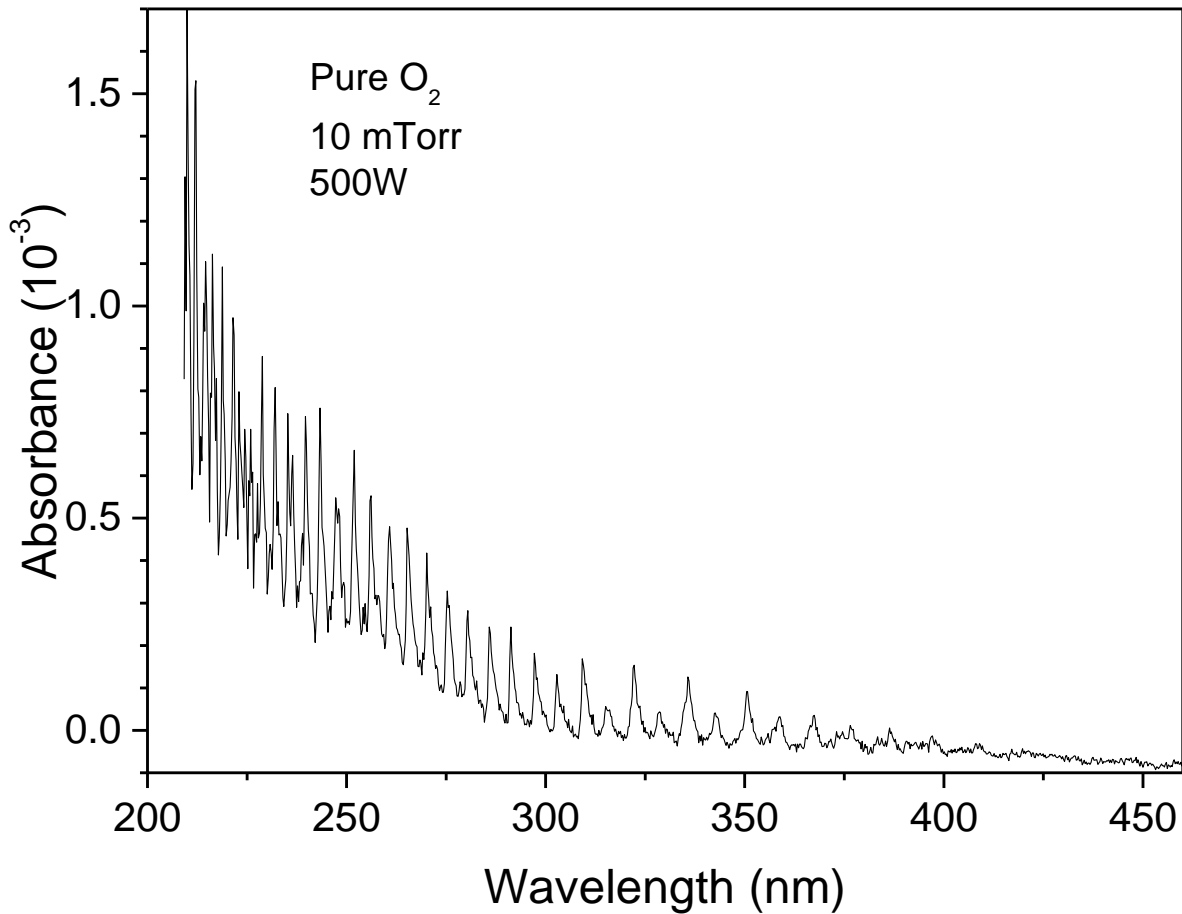
- Baseline noise $\approx 10^{-5}$, 250nm spectrum simultaneously
- Allows whole vibrational bands to be observed

Pure O₂ plasma UV absorption



Cold O₂ doesn't absorb
above ≈ 200 nm...

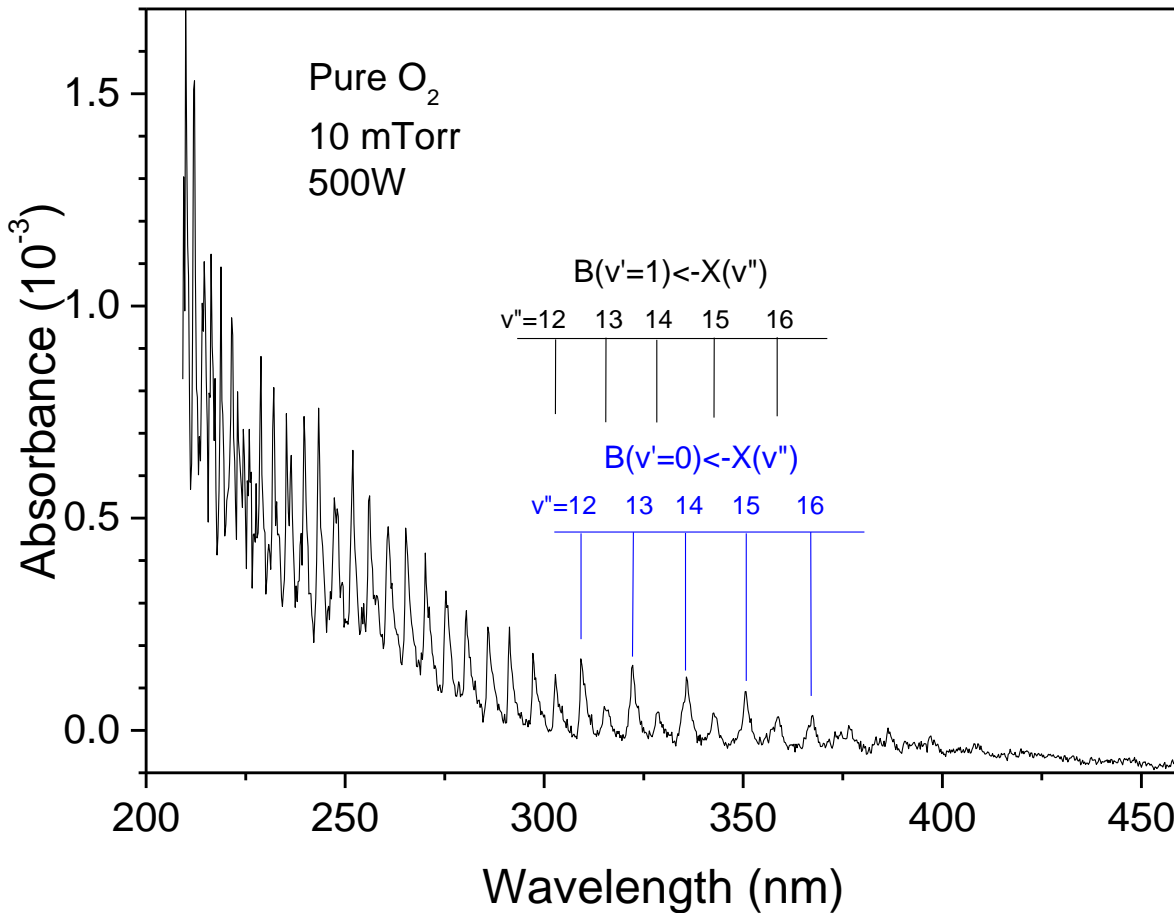
Pure O₂ plasma UV absorption



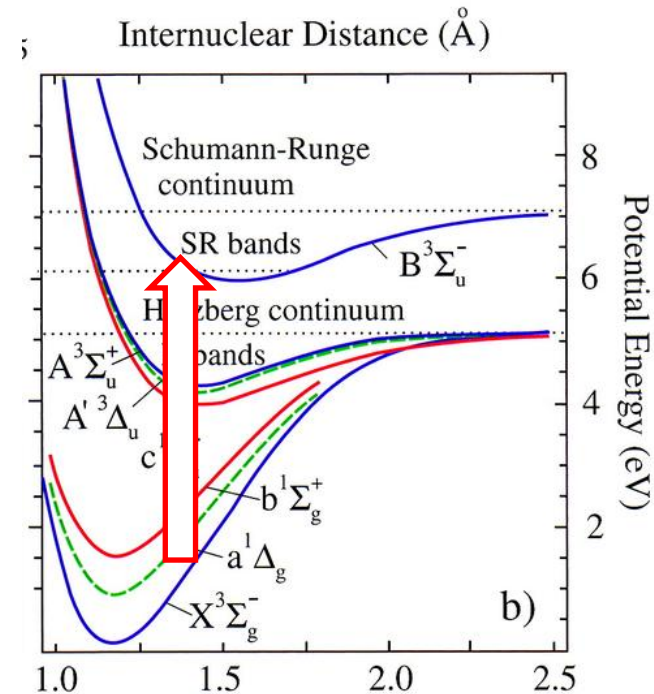
Pure O₂ plasma UV absorption



Levels up to $v'' \approx 18$ half-way to dissociation!



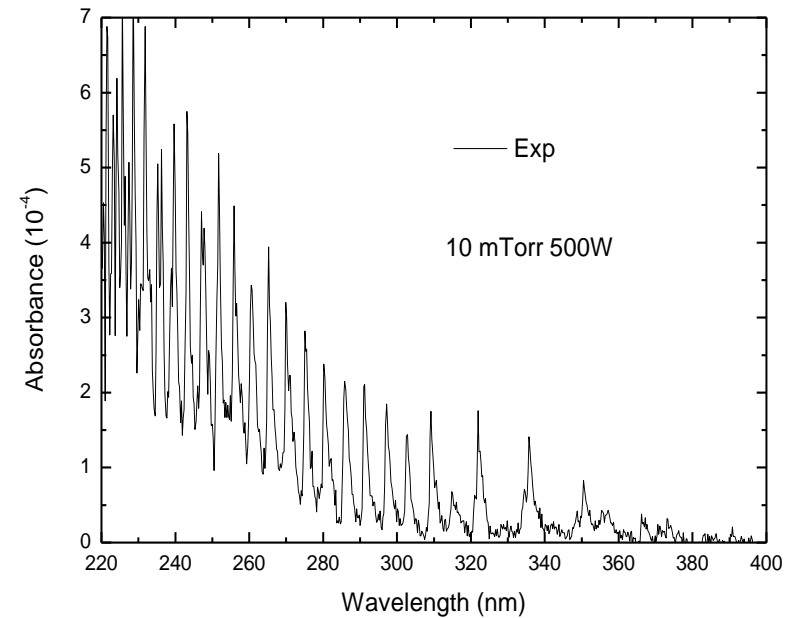
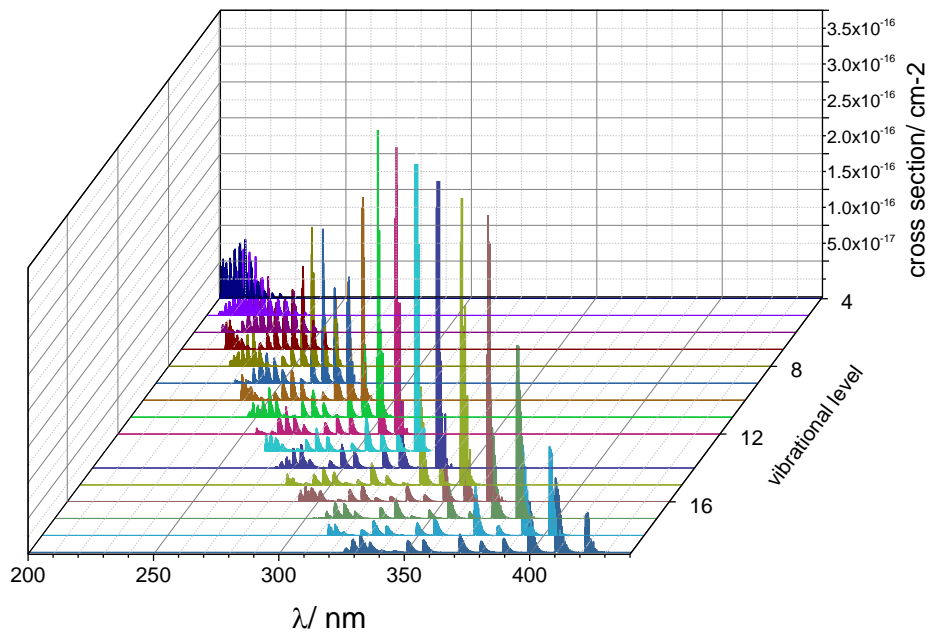
Schumann-Runge bands
from highly vibrationally
excited O₂



Pure O₂ : UV absorption



Vibrational-state resolved absorption cross-sections
→ Extract vibrational distribution functions

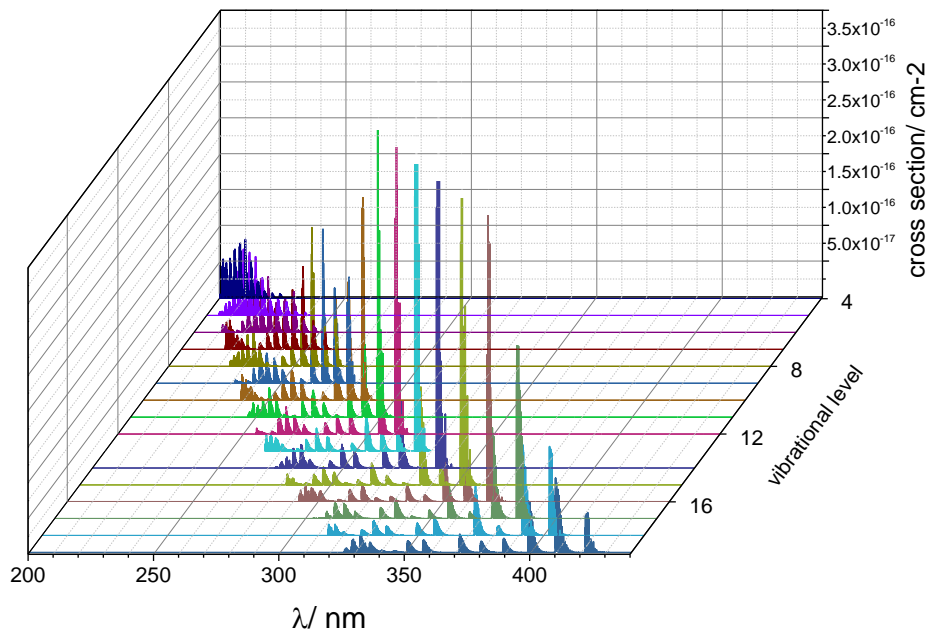


(Courtesy of Christophe Laux / Specair)

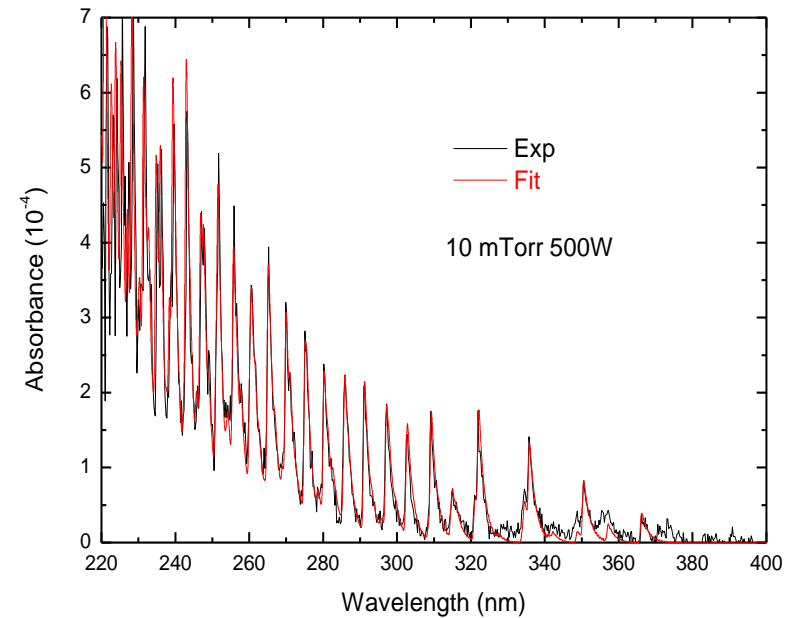
Pure O₂ : UV absorption



Vibrational-state resolved absorption cross-sections
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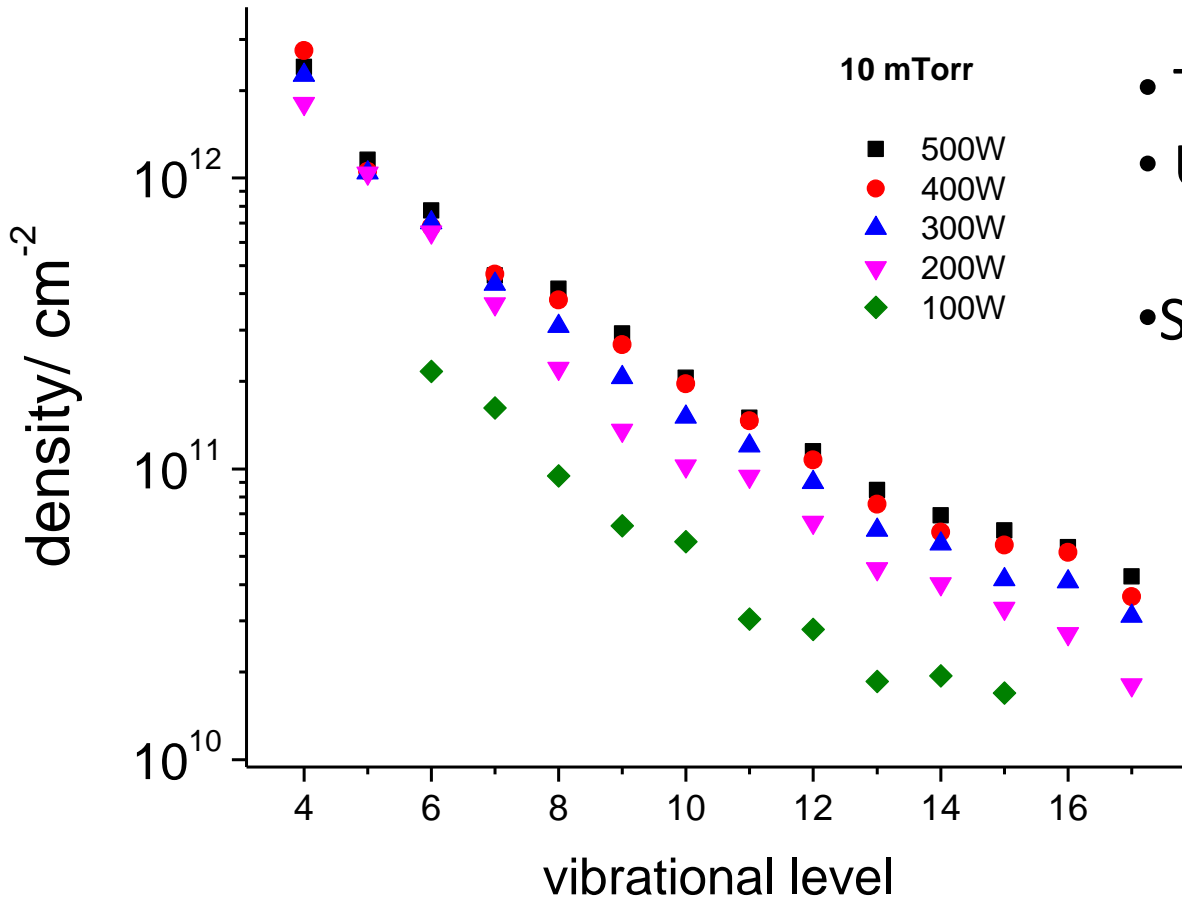


(Courtesy of Christophe Laux / Specair)



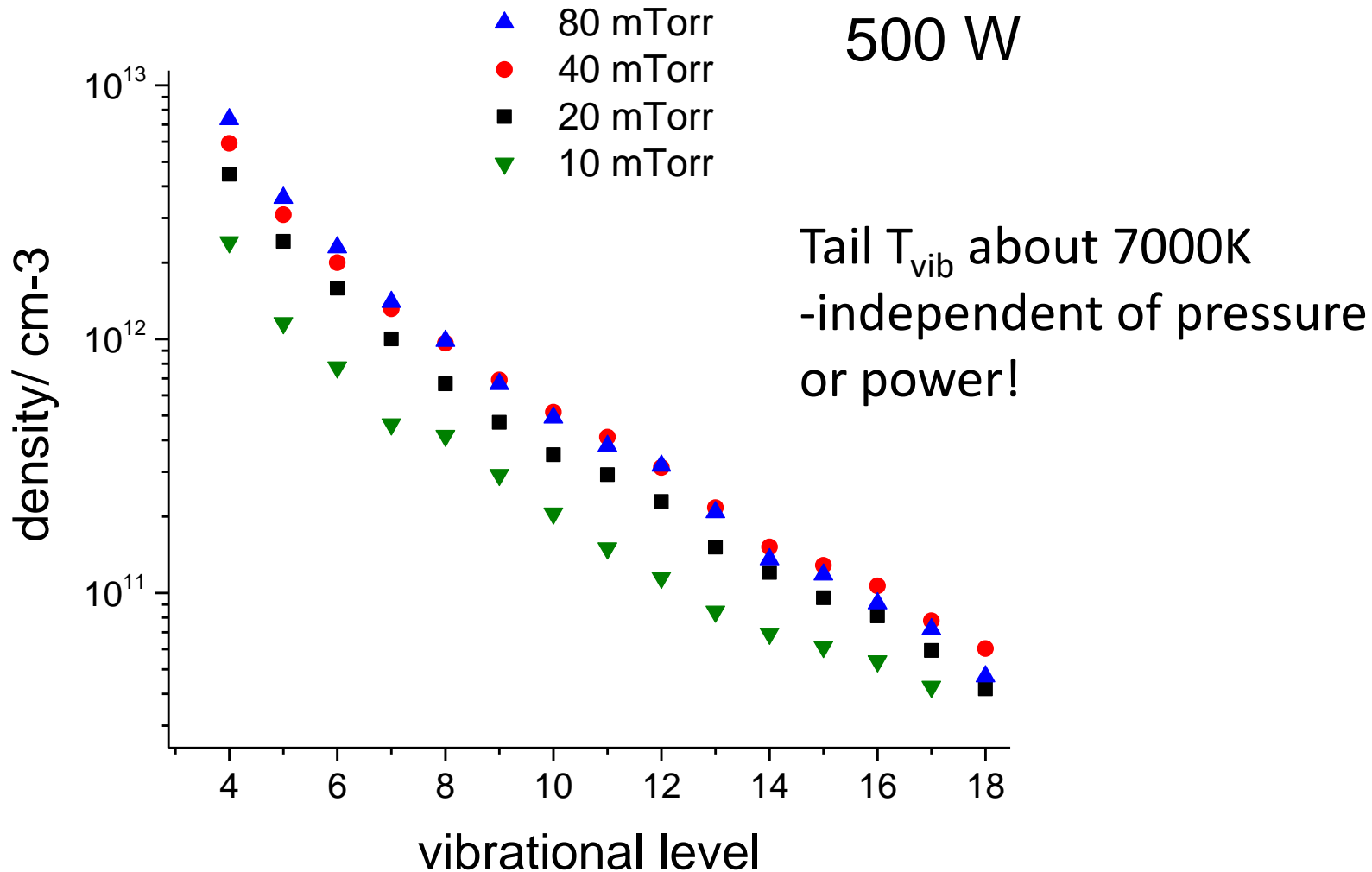
(NB cannot measure below $v''=4$)

O₂ Vibrational distribution functions (function of RF power :10 mTorr)

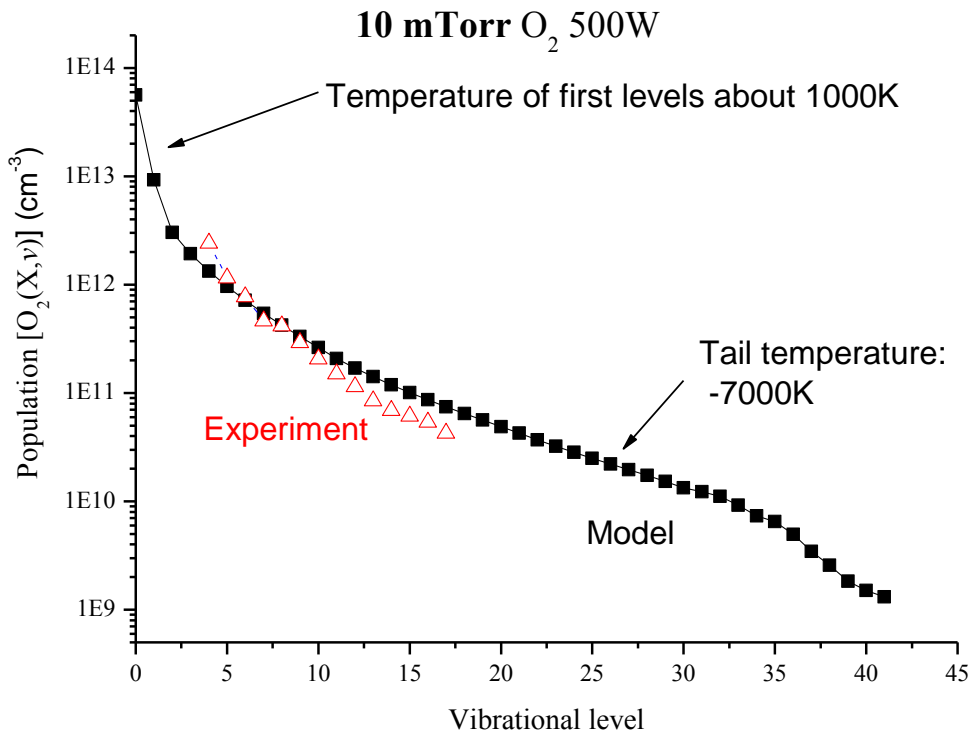


- Tail $T_v \approx 7000$ K
- Up to 3% of O₂ in $v = 4-18$
- Saturates at high power

O₂ Vibrational distribution functions (function of pressure: 500W)



Modelling the VDF: IST Lisbon



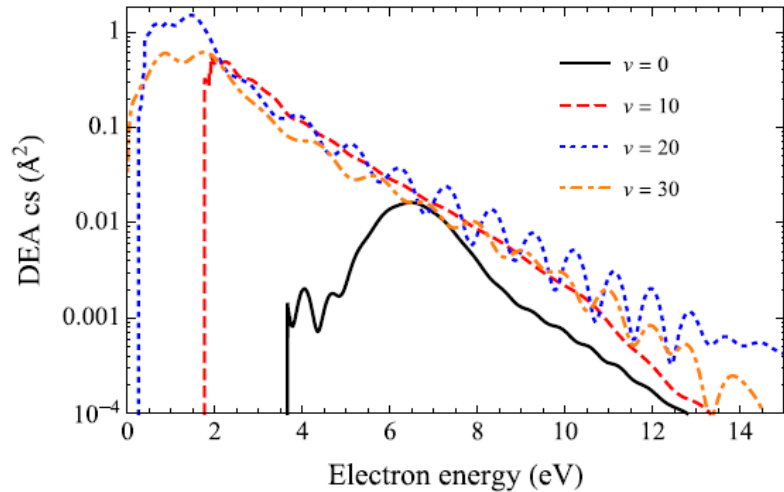
Lowest levels in equilibrium with T_{gas}

Higher levels much hotter

VDF determined by:

- electron impact excitation
- V-T O₂ - O

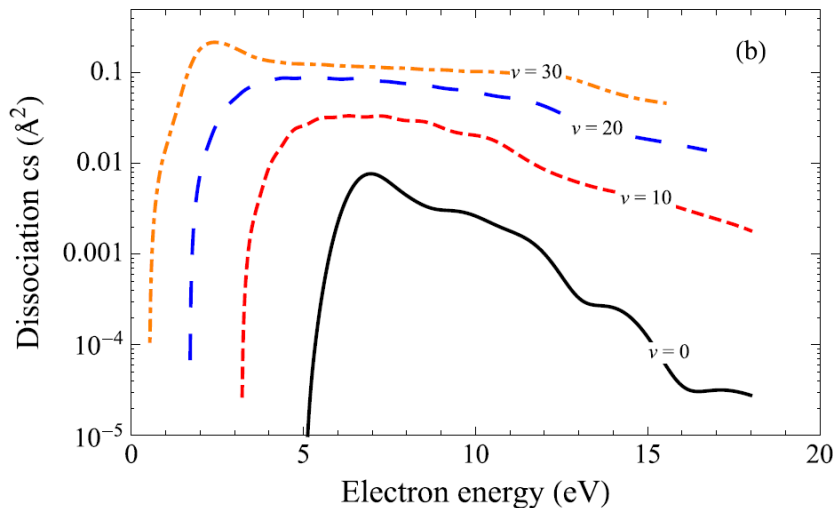
Effect of vibrational excitation on electron-induced processes



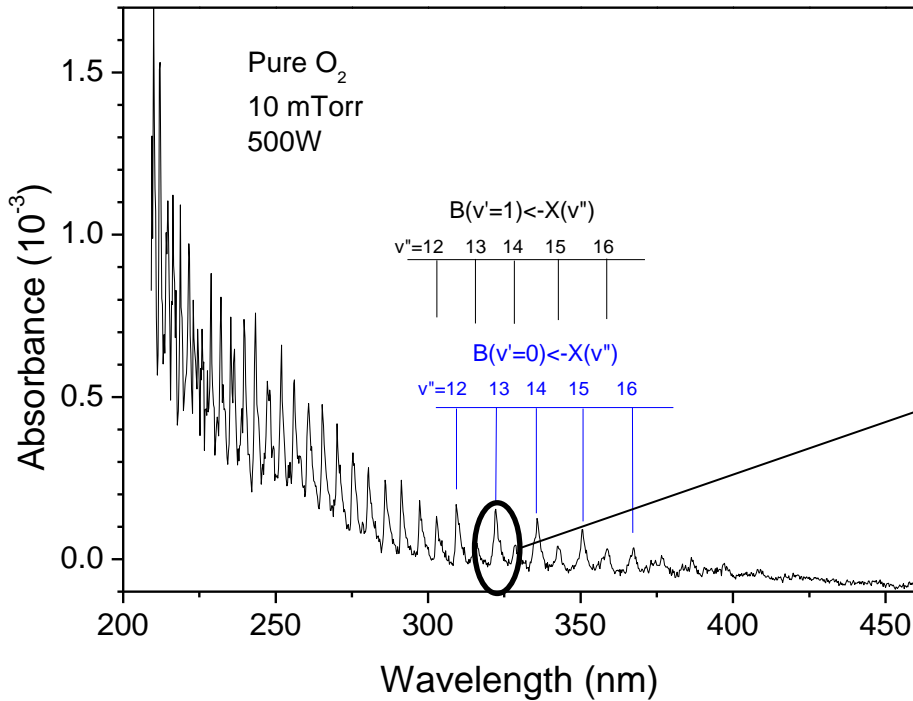
Excitation to $v > 10$ has a strong effect on both cross-sections:

- Lower threshold
- Higher cross-section

Dissociation and negative ion production significantly enhanced

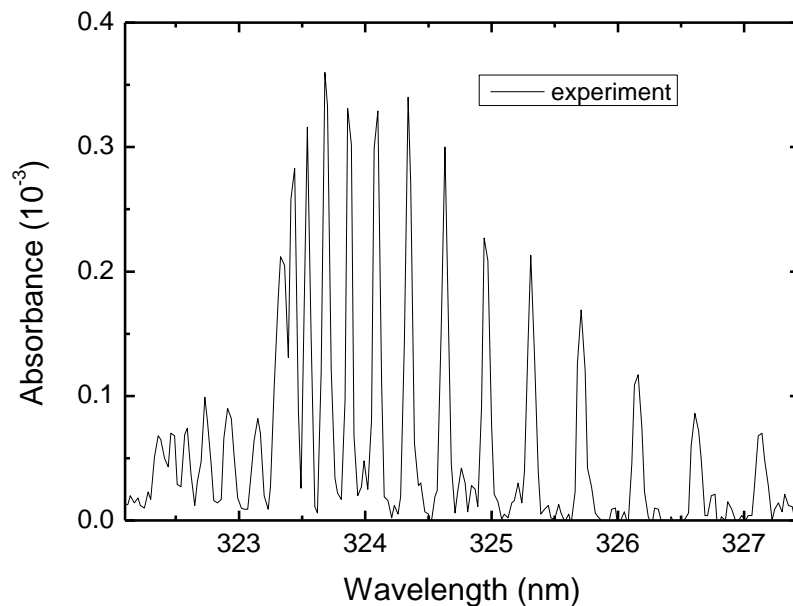
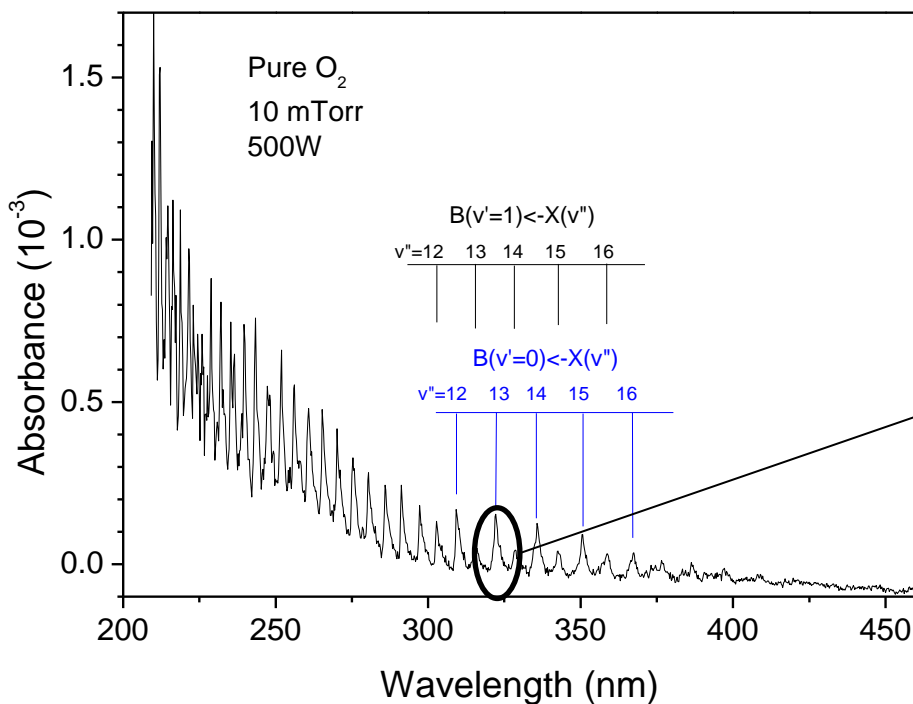


Gas temperature: Absorption spectra at higher resolution:



Look at one band in higher resolution :
Change grating 300l/mm to 2400l/mm:

Absorption spectra at higher resolution:



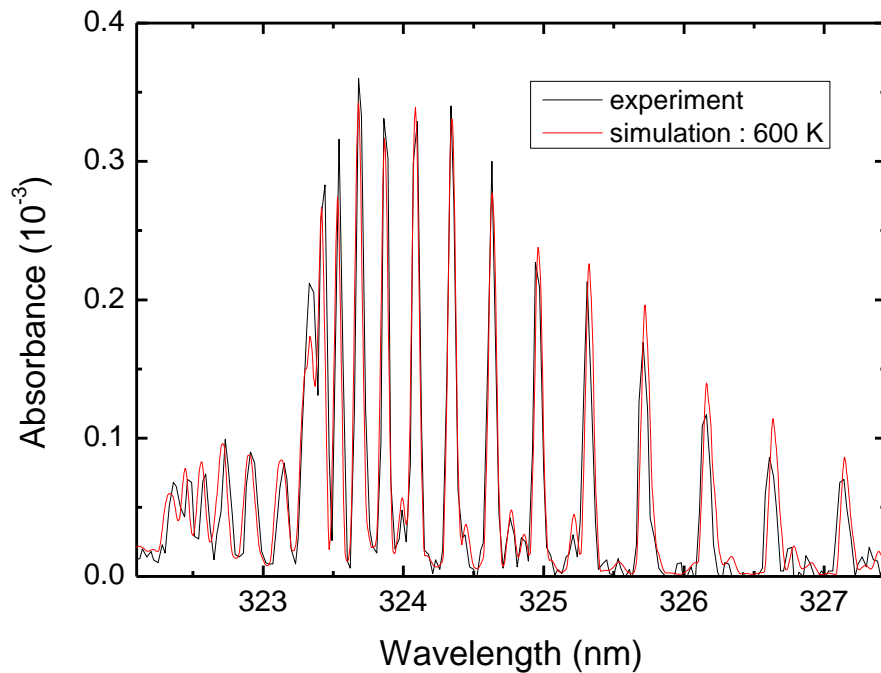
Observe rotational structure

At these pressures $T_{\text{trans}} \approx T_{\text{rot}} \ll T_{\text{vib}}$

Absorption spectra at higher resolution:



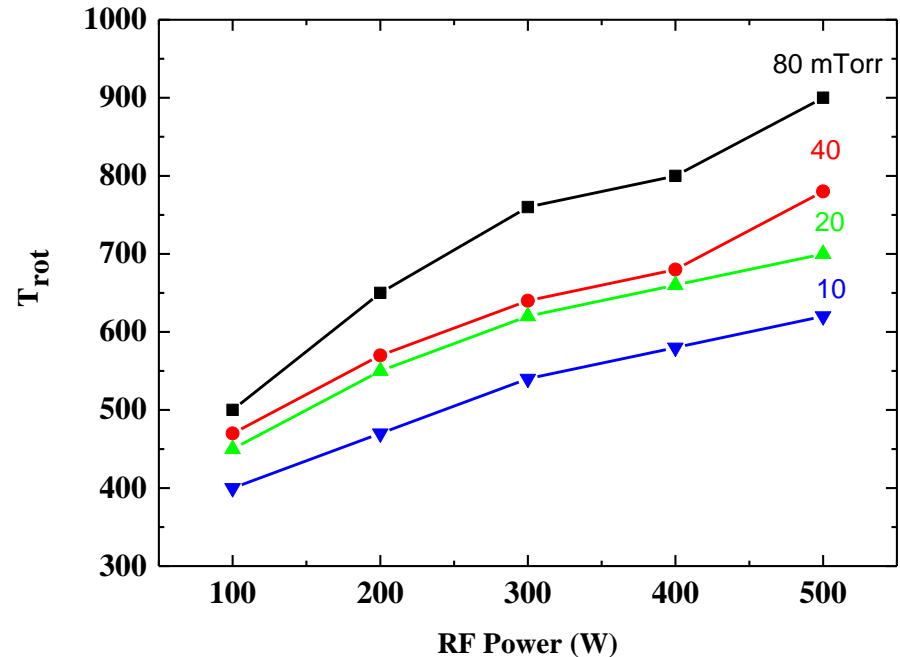
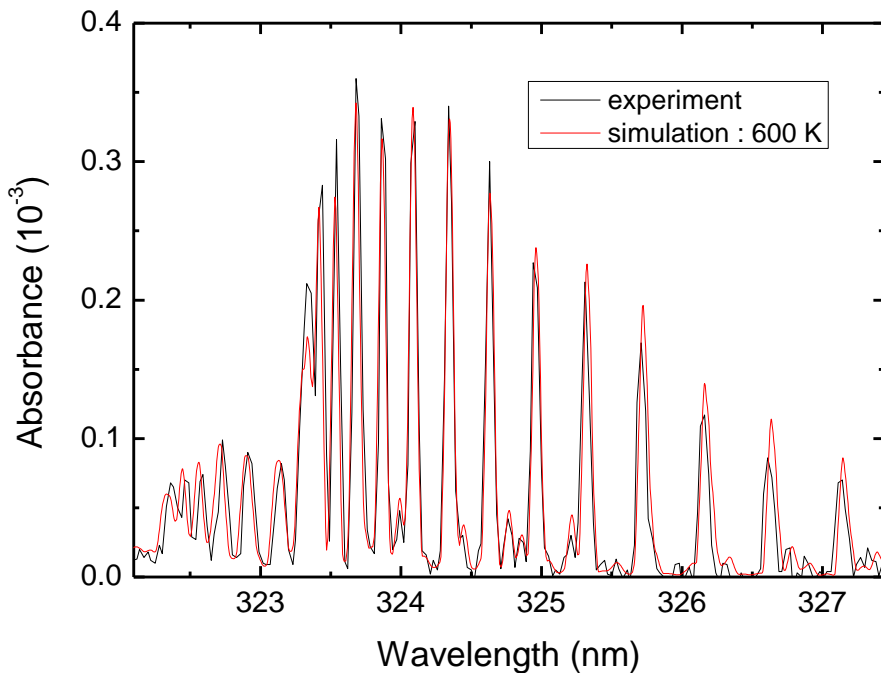
Fit to simulated spectra to determine T_{rot} :



O₂ rotational temperature



Fit to simulated spectra to determine T_{rot} :



T_{rot} up to 900K!

Explains why O atom density does not exceed 30% of initial gas density

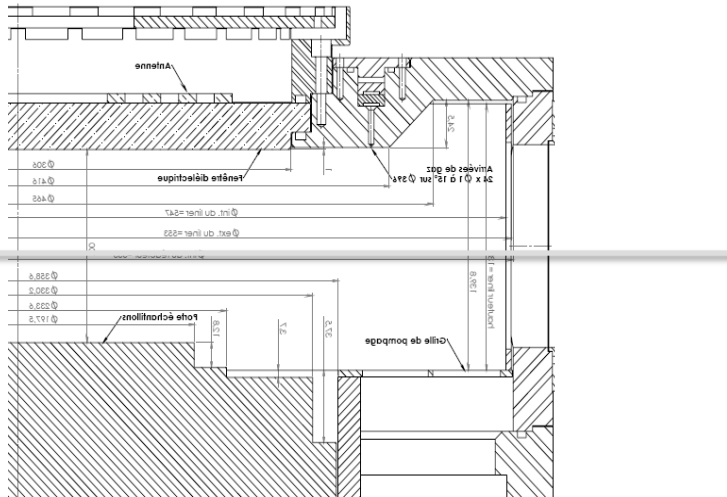
Gas temperature from Ar^m IRLAS Doppler width



Add 10% Ar

Determine temperature from
Doppler width of Ar metastable
absorption at 772nm

Laser beam at reactor mid-plane:

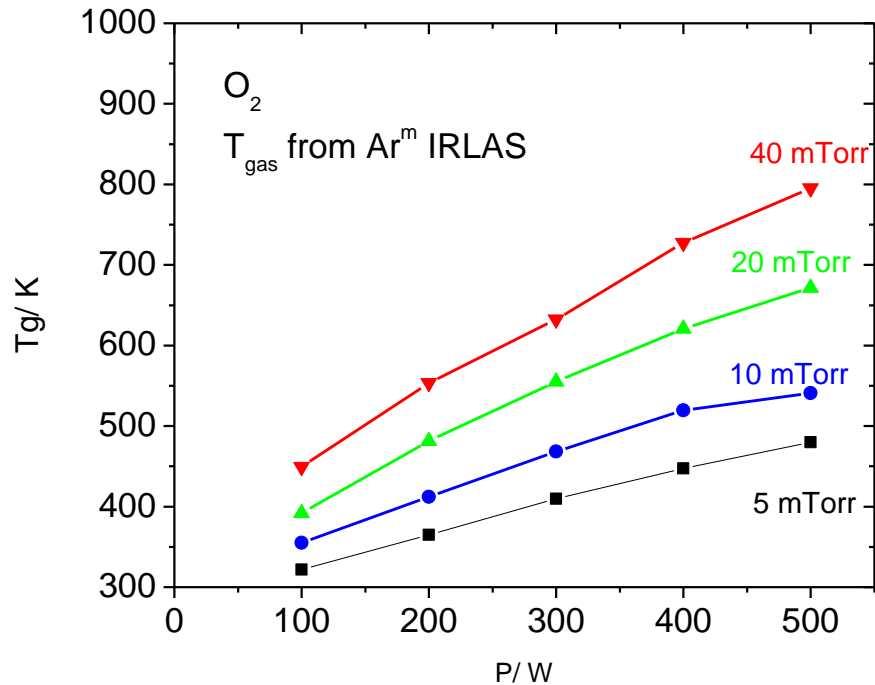


NB measurement integrated over the
reactor diameter, weighted by the Ar^m
density profile

Gas temperature from Ar^m IRLAS Doppler width



From Ar^m IRLAS

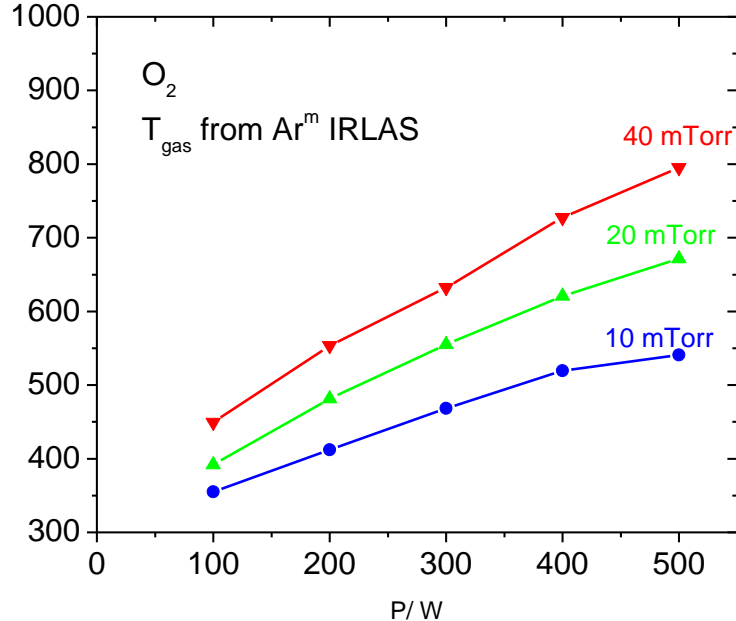


Is thermal equilibrium $T_{\text{trans}} = T_{\text{rot}}$ established at the lowest pressures?

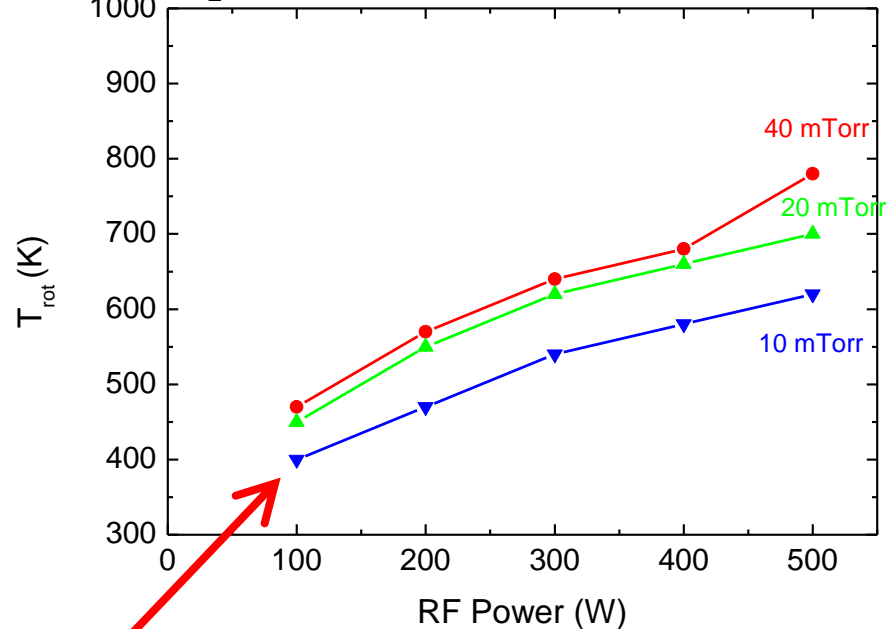
Compare the two techniques:



Ar^m Doppler temperature;



O₂ rotational temperature:



Reasonable qualitative agreement

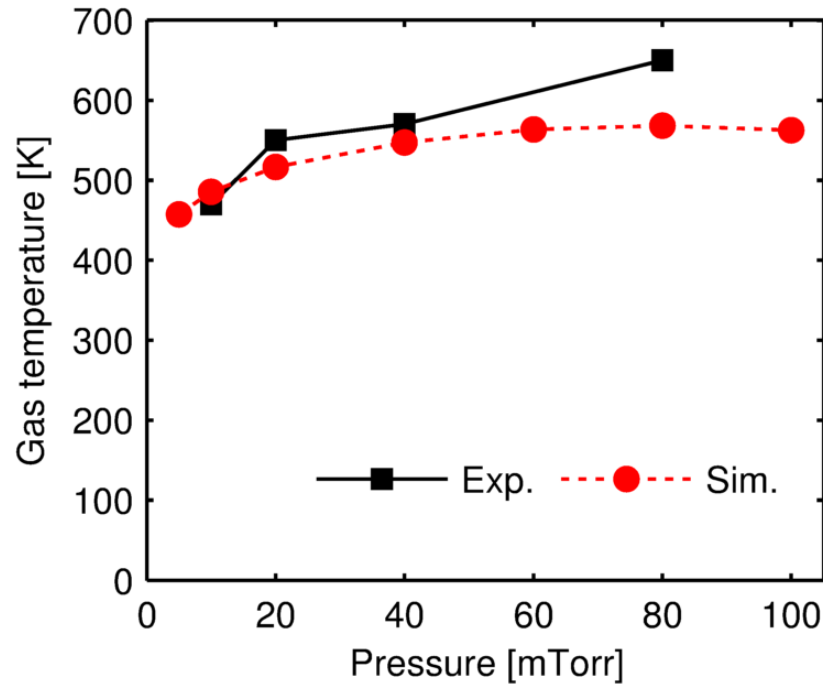
-but at low pressure, IRLAS is cooler than O₂ T_{rot}

Can we directly measure translational temperature of O atoms?

-with time and space resolution

-probe energy relaxation rates (surface and gas phase)?

Simulated gas temperature (300W)



Reasonable trend but underestimated heating:

-underestimated energy release from dissociation?

-neglected heating mechanisms?

-overestimated thermal accommodation at walls?

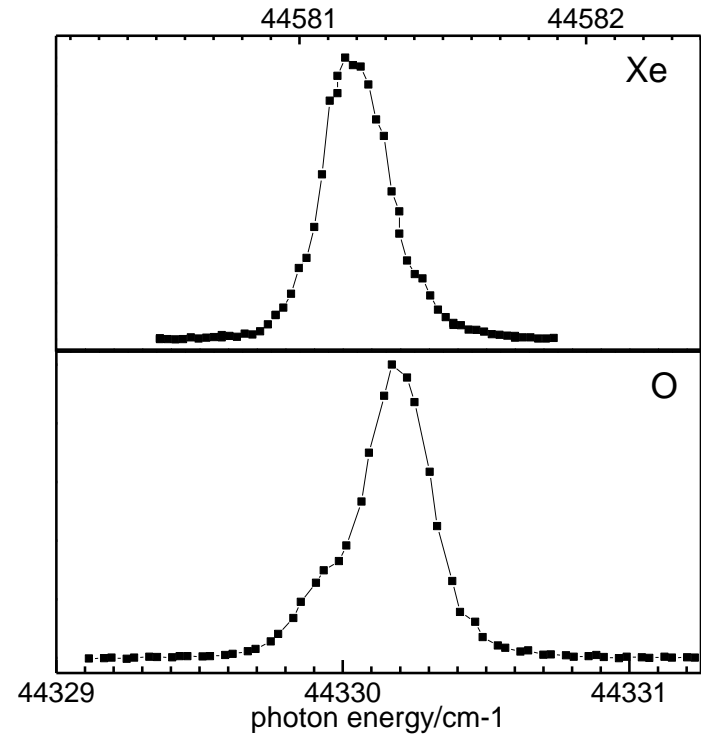
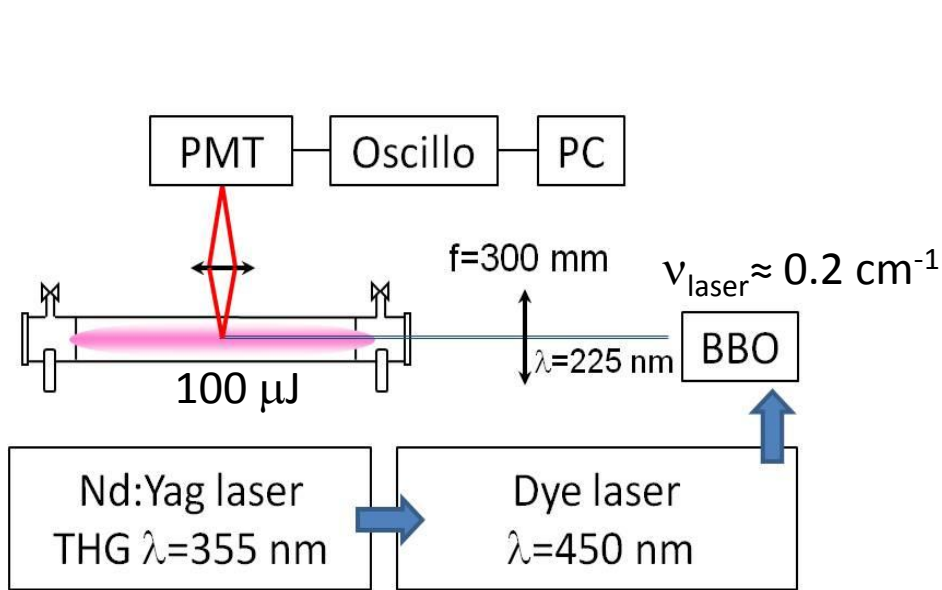
High resolution TALIF for 0 atom temperature



D. Marinov, O. Guaitella, M. Foucher, JP. Booth
(LPP)

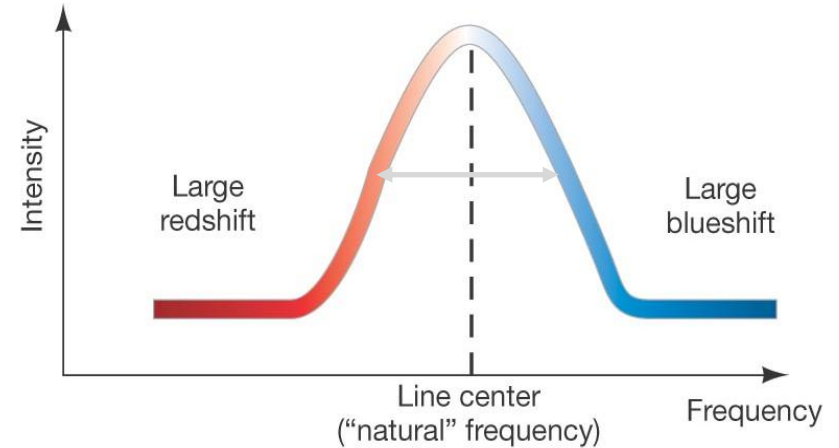
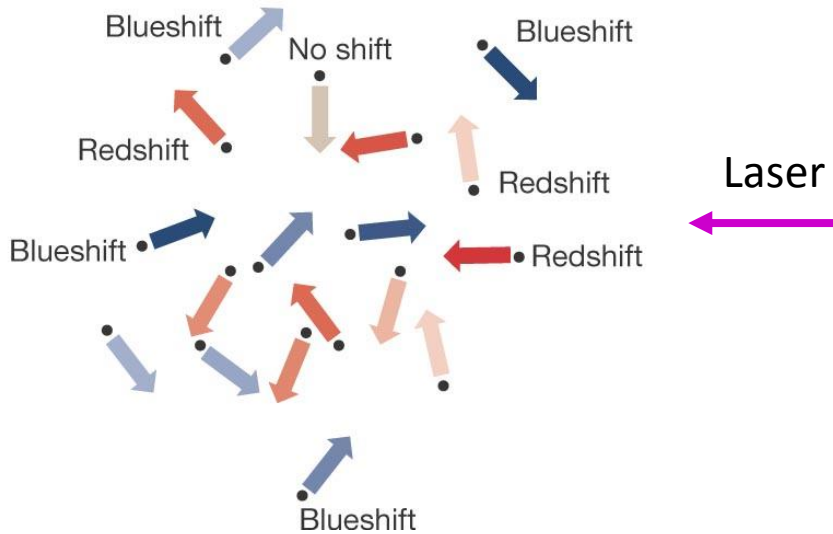
C. Drag, C. Blondel
(Lab Aime Cotton)

Standard TALIF



- A lot of information is hidden in the line shape!
- But it is not accessible because of the broad and often unknown laser line profile.

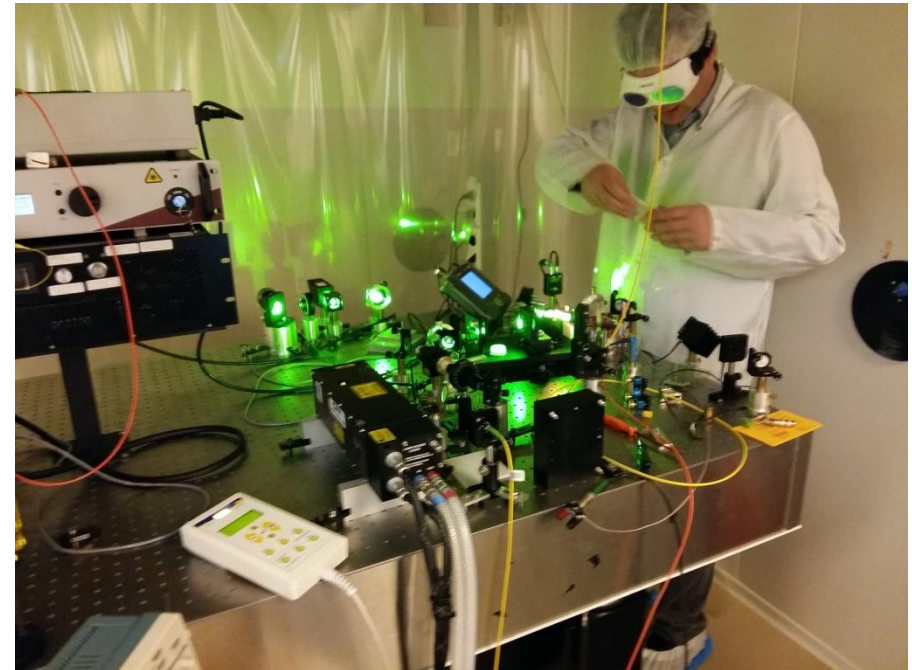
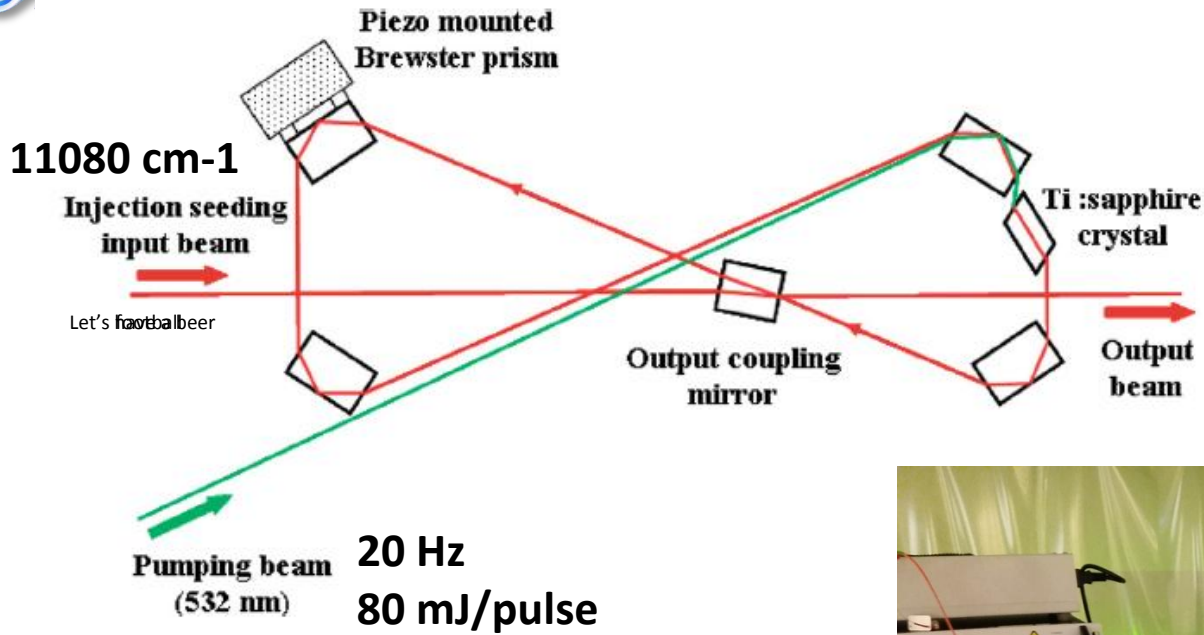
Doppler line width – a direct measure of the **translational** gas temperature



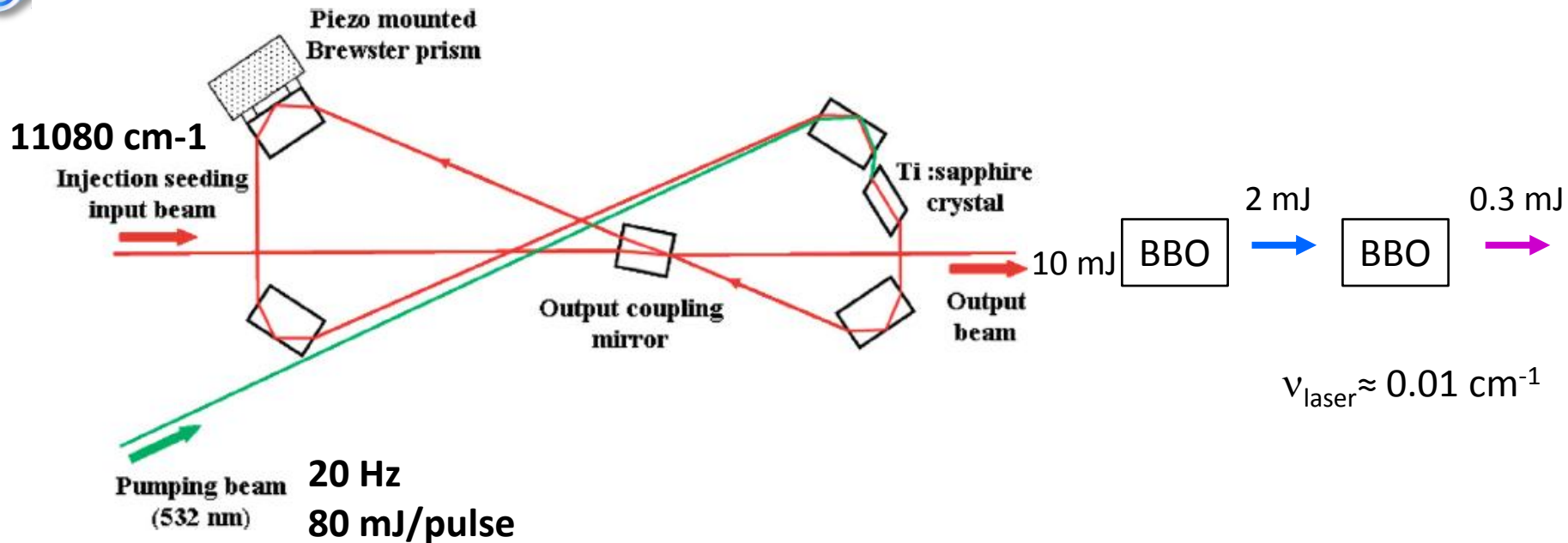
$$\frac{\Delta \nu_{FWHM}}{\nu_0} = \sqrt{\frac{8kT \ln 2}{Mc^2}}$$

$$\text{At } 300 \text{ K } \Delta \nu_0 = 0.3 \text{ cm}^{-1}$$

Single mode pulsed laser (Aimé Cotton)



Single mode pulsed laser (Aimé Cotton)



- Doppler and sub-Doppler line profiles can be accurately measured.

Laser performance



Fundamental line width: determined by Fourier transform of pulse duration (22MHz) + locking jitter (25MHz)

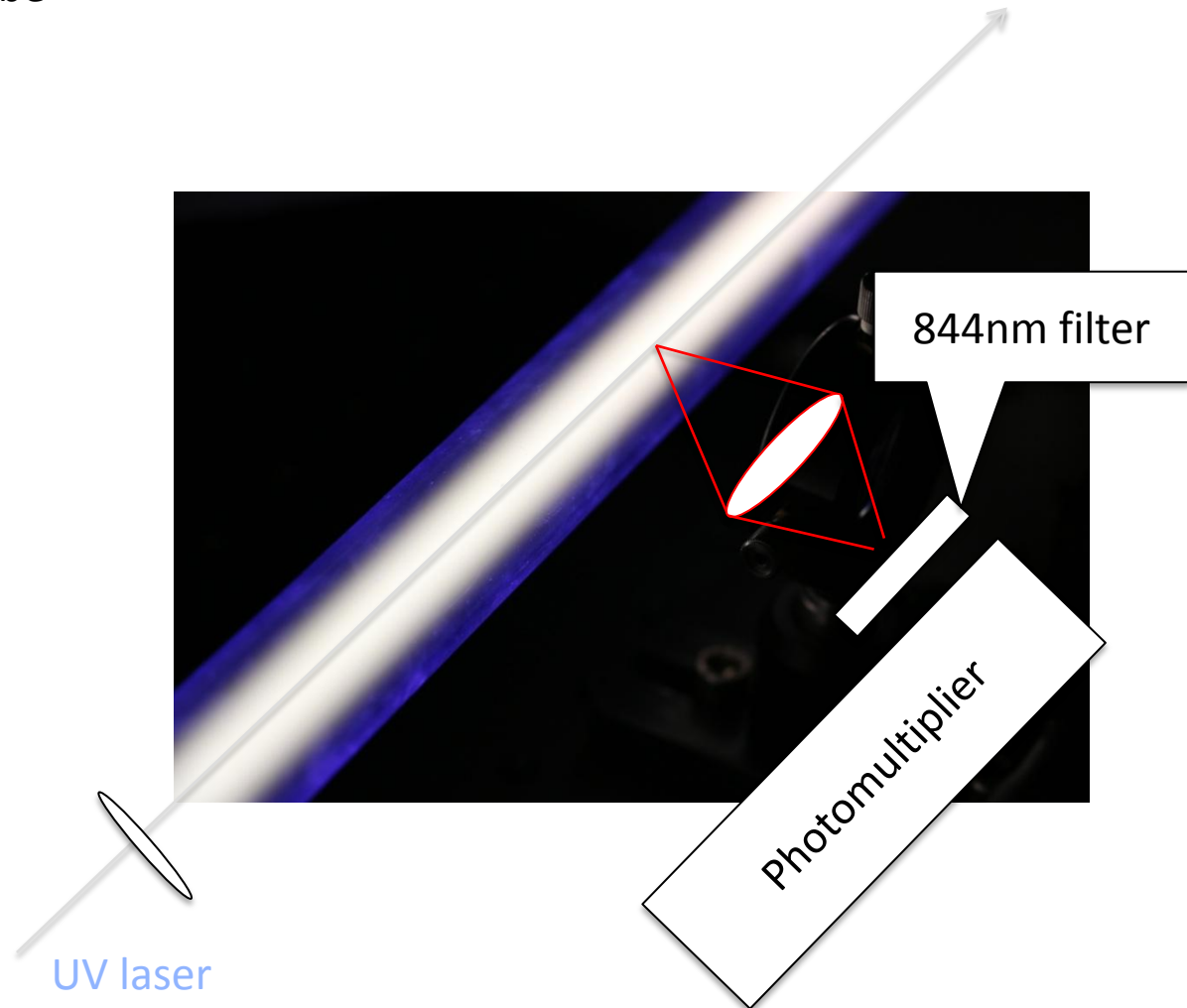
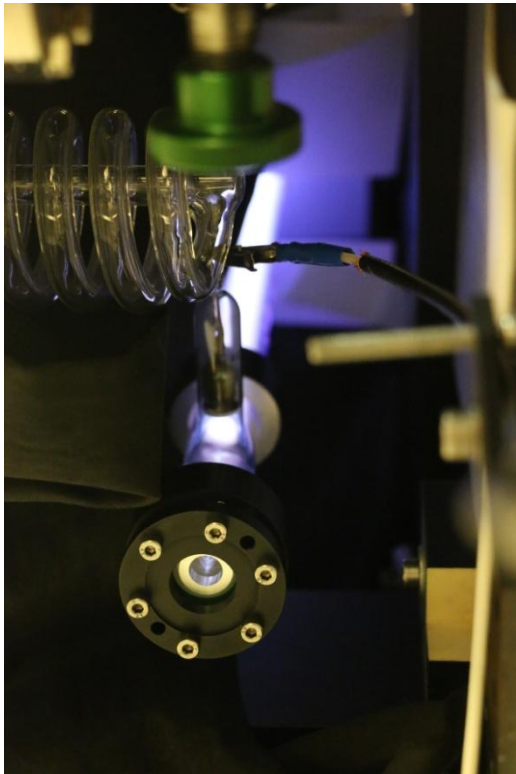
	Wavelength (nm)	Laser energy (mJ)	Estimated line width
Fundamental	902.32	10	47 MHz
Frequency doubled	451.16	2	
Frequency quadrupled	225.58	0.25	188 MHz (0.01cm^{-1} @ 2 photon)

30x narrower
than a dye laser!

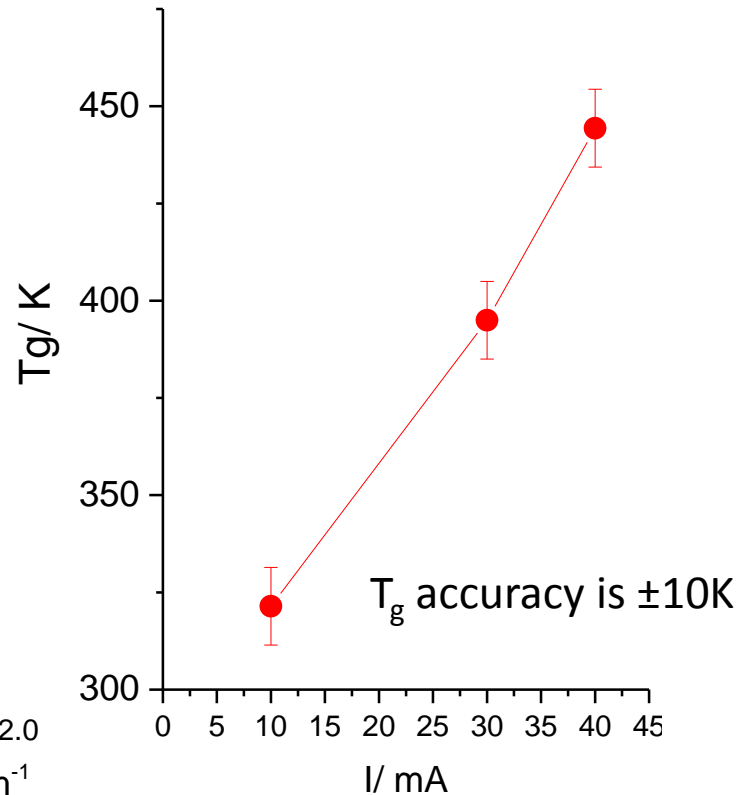
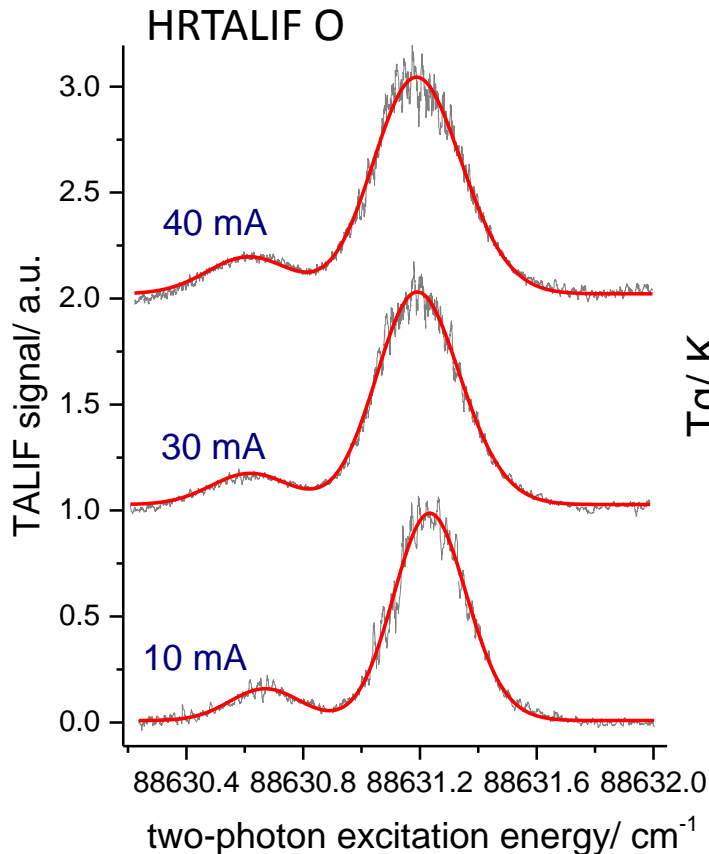
O atoms in an O₂ DC discharge



1-4 Torr O₂ in a 2cm diameter tube
5-40 mA current

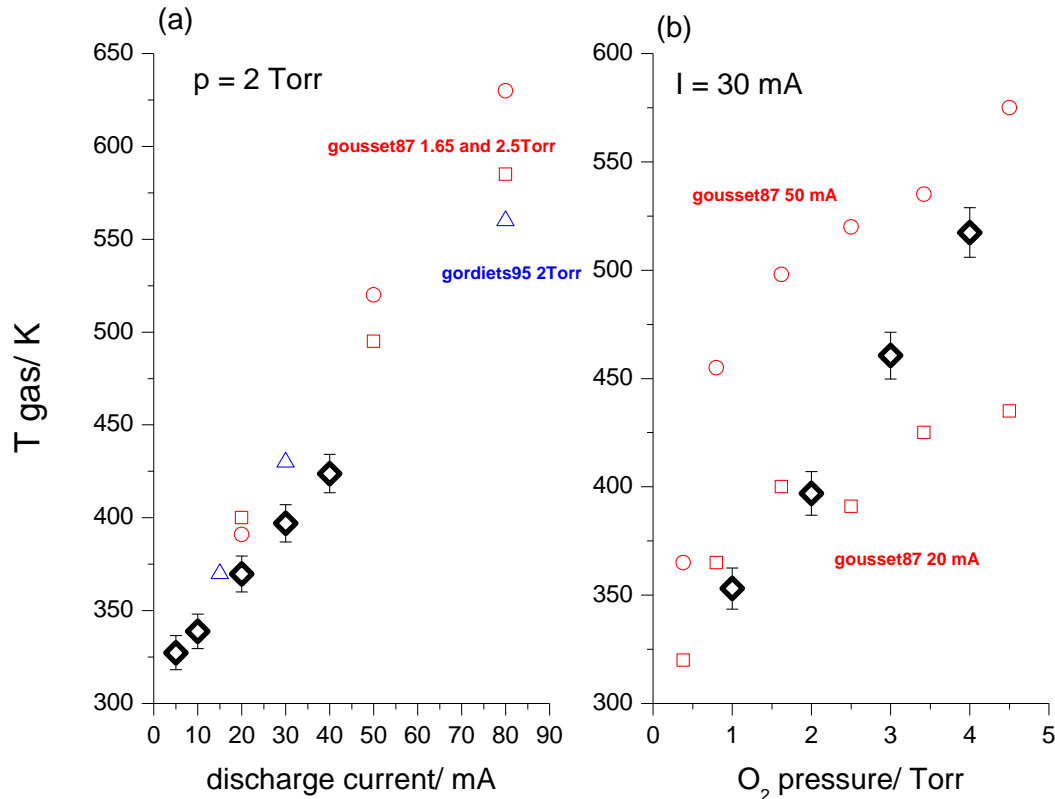


Gas temperature measurements in a low pressure dc glow discharge



O_2 dc discharge 2 Torr – collisional broadening is negligible

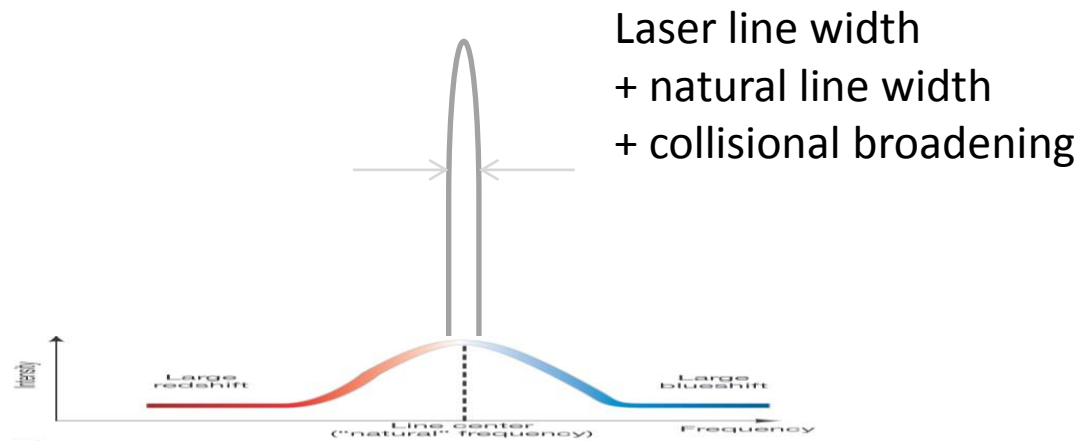
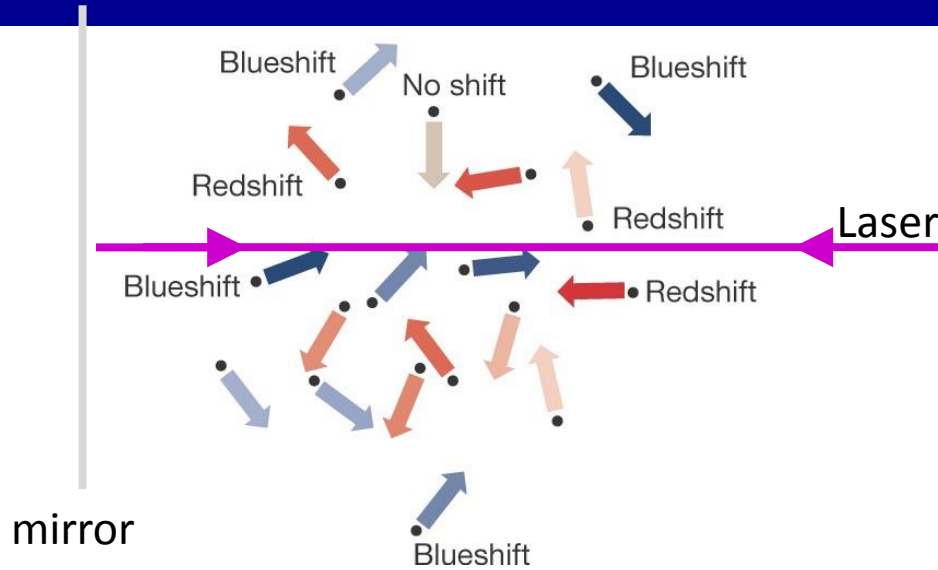
Comparison to previous measurements



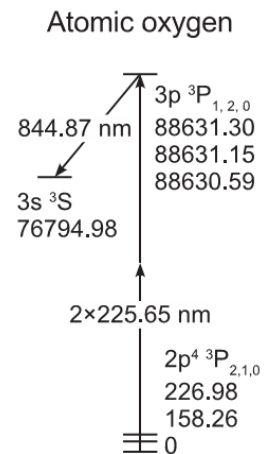
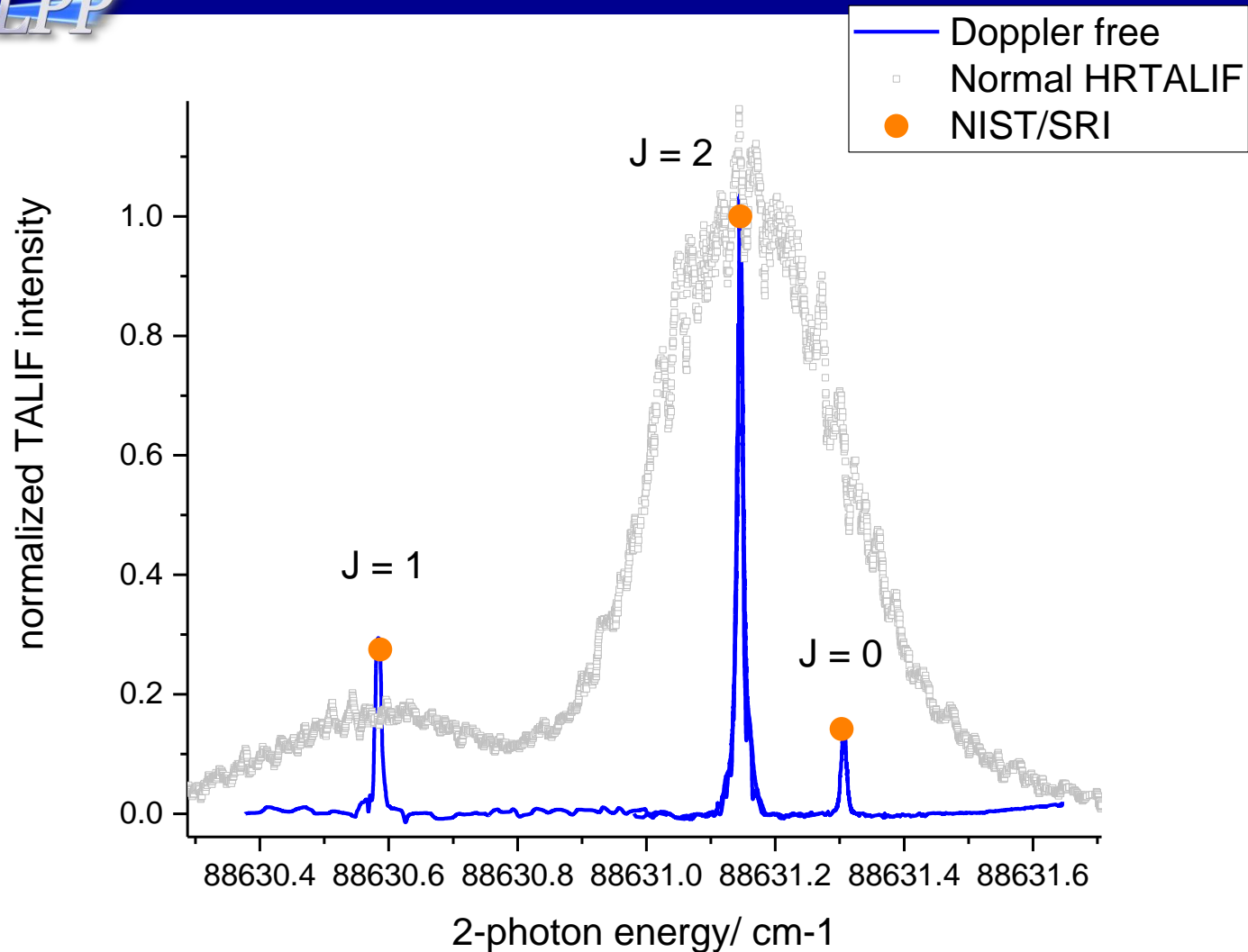
(T deduced from O_2 density :
VUV absorption)

Good agreement within
errors of previous
measurements

Doppler-free TALIF spectroscopy

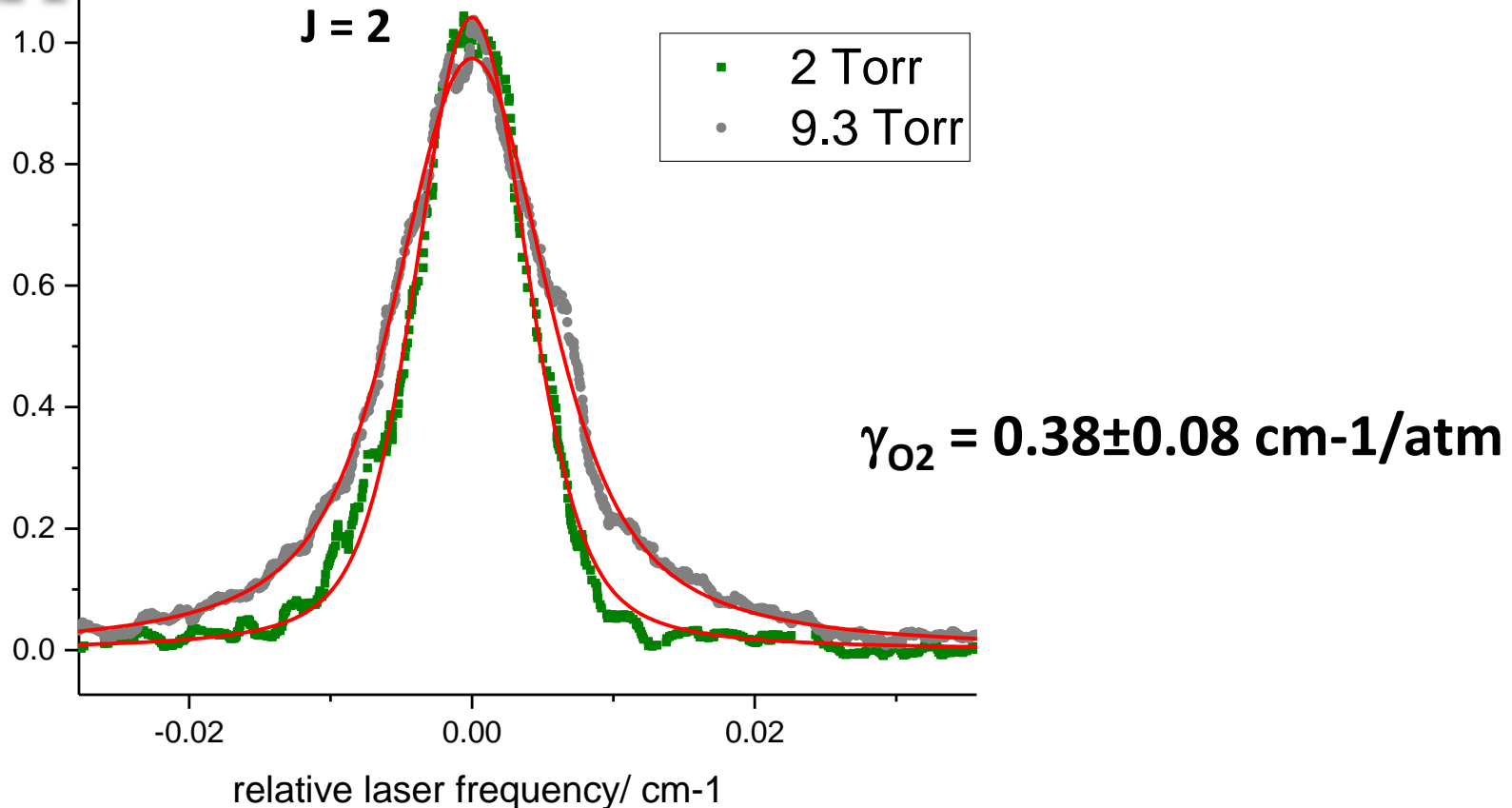


Doppler-free measurements of the fine structure components of O^3P



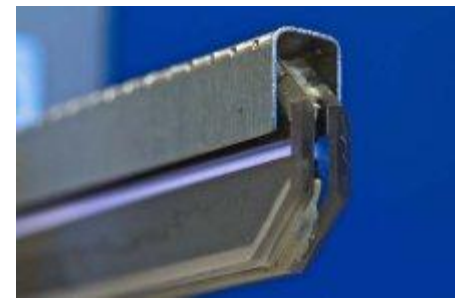
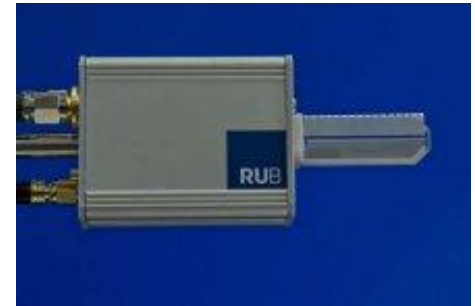
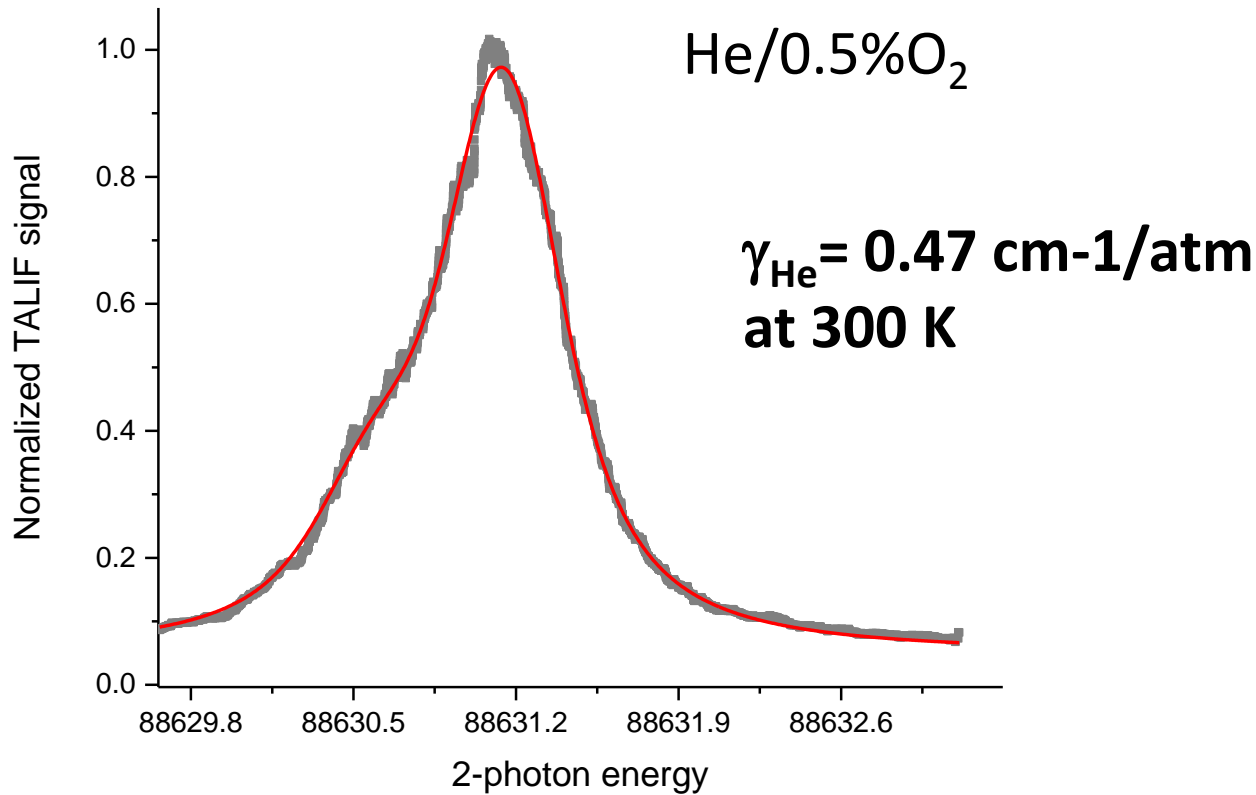
O_2 dc discharge 2 Torr

Measurements of the collisional broadening coefficients in the Doppler-free configuration

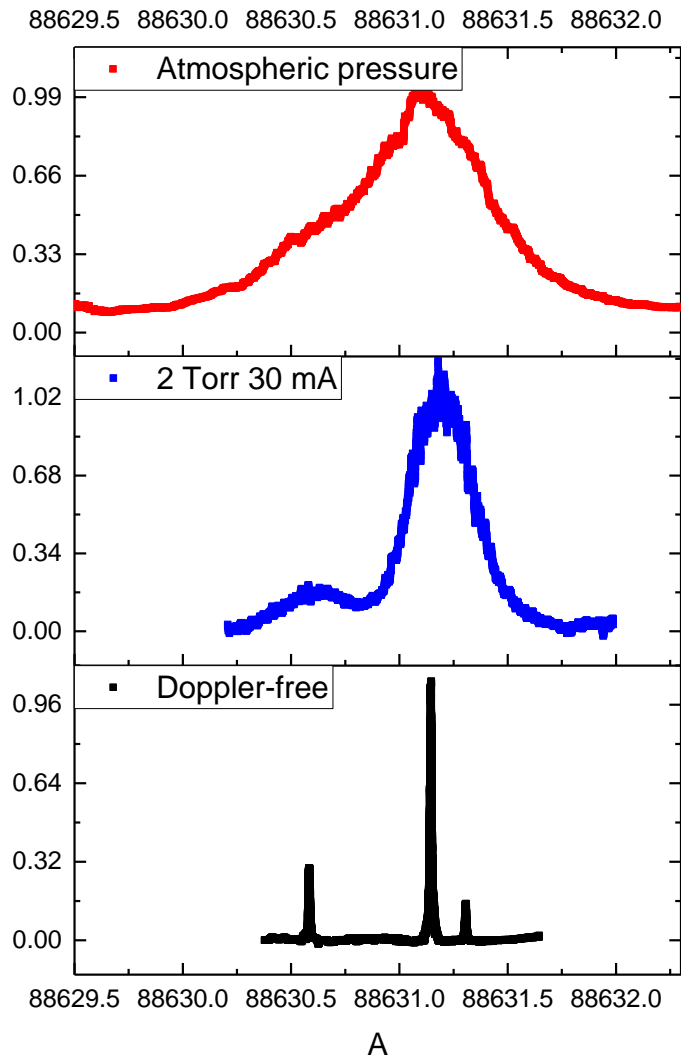


The only literature value¹ $\gamma = 0.42 \text{ cm}^{-1}/\text{atm}$ in a O_2/CH_4 flame at 2500 K
¹Dyer et al., Opt. Lett. **14**, 12-14 (1989)

Pressure broadening in the μ APPJ (normal TALIF configuration)



Summary of HR TALIF



- atm pressure:
 - Pressure broadening
- Low pressure Doppler:
 - Translational gas temperature
- low pressure sub-Doppler
 - Pressure broadening
 - Fine structure
 - Atomic physics

Conclusions



Comprehensive data set to test models of Oxygen plasmas

- Atom densities and kinetics
- Molecule densities and energy distributions
- Electron (& negative ion) densities,
- Gas temperature

Even for the simple case of pure diatomic gases, **state-of-the-art models are unable to correctly predict trends with gas pressure**

- Fundamental collision data is lacking or innaccurate**
- Gas heating** is very significant and remains to be fully understood
- Vibrational excitation** can be significant and may play a large role

Future work:

High resolution TALIF – in ICP reactor

- time and space resolution to understand relaxation kinetics

FUNDING :

