

# Errors and uncertainty in oxygen chemistry

Miles M. Turner

Dublin City University, Ireland



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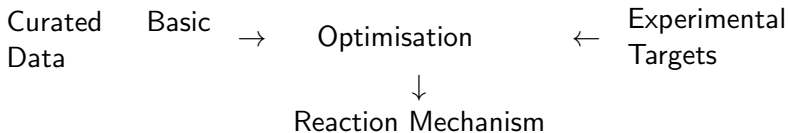
# Programme

- Best practice in plasma chemistry modelling: “Reaction mechanisms”
- Tools: Uncertainty quantification, sensitivity analysis, dominant pathways
- Oxygen chemistry: Outcomes

## Best practice? “Reaction Mechanisms”

- A “reaction mechanism” is a chemistry model with these features:
  - ① Based on “curated” basic data  
(Sources of data identified, error bars (uncertainty) associated with data, multiple/contradictory sources critically evaluated to “best” value, *etc*)
  - ② Validated against experimental “targets”:  
(Some set of experiments to be reproduced, also critically evaluated, with error bars, *etc*)
  - ③ Optimized against “targets”:  
(Uncertain rate constants systematically adjusted to minimise disagreement with targets.)
- Example:  
GRI Mech 3.0, optimised natural gas combustion model with  
~ 325 reactions, ~ 50 targets

# Developing a “Reaction Mechanism”



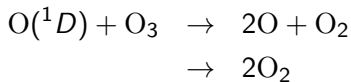
- Changes to basic data or targets trigger a reoptimisation
- Tinkering with an optimised model is unlikely to be a good idea

## Present situation

- Our older literature contains examples of very good practice (by the standards of the time):  
Gordiets, *et al*, "Kinetic model of a low-pressure N<sub>2</sub>-O<sub>2</sub> flowing glow discharge," J. Phys D 23, 750 (1995)
- But we don't seem to have improved since, and arguably there has been decline

## Need for Curation: Quality of Data

- The reaction



is the subject of about 10 experimental studies and three critical reviews (1987,2004,2011)

- The critical recommendation is

$$k = 2.4 \times 10^{-16} \text{ m}^3 \text{ s}^{-1}$$

with equal branching  
Established for almost 30 years!

- A look at nine models featuring this reaction shows:

- ① None cites a critical review as authority
- ② Seven have the wrong rate constant and/or branching ratio
- ③ One has a rate constant almost 5 times too large

- Why?  
Misunderstanding complex sources, uncritical copying, unclear referencing

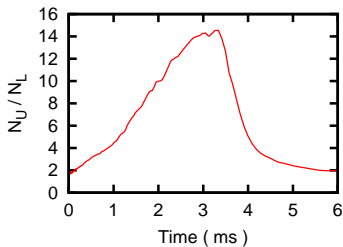
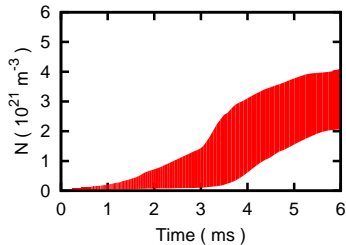
# Oxygen Plasma Chemistry

- Oxygen is clearly an important chemistry  
Perhaps dozens of species, hundreds of reactions, all with uncertain rate constants
- Models aim to predict (?), but predictive power is compromised by uncertain rate constants
- The predictive uncertainty involved can be large  
MMT, Plasma Source Sci. Technol. 24, 035027 (2015)
- What is an appropriate procedure for dealing with such models?

# Uncertainty by Monte Carlo simulation

- He/O<sub>2</sub> chemistry, 373 reactions, 25 species
- Each rate constant has an error bar:  $k \pm \Delta k$
- Monte Carlo procedure maps uncertainty in rates to uncertainty in density:

$$k \pm \Delta k \xrightarrow{\text{Monte Carlo}} n \pm \Delta n$$





## Sensitivity Analysis

- Which rate constants cause uncertainty?
- “Sensitivity analysis” aims to answer this question
- Basic concept:
  - 1 Isolate effects by changing a single rate constant in each trial (applying an “elementary effect”)
  - 2 Obtain a global picture by averaging many such effects

So

$$\begin{pmatrix} k_1 \\ k_2 \\ \vdots \\ k_i + \Delta k_i \\ \vdots \\ k_N \end{pmatrix} \rightarrow \begin{pmatrix} n_1 + \Delta n_{1,i} \\ n_2 + \Delta n_{2,i} \\ \vdots \\ n_j + \Delta n_{j,i} \\ \vdots \\ n_M + \Delta n_{M,i} \end{pmatrix}$$

and

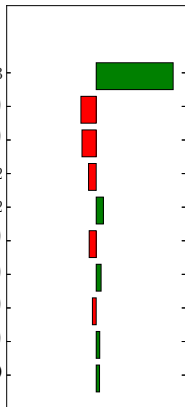
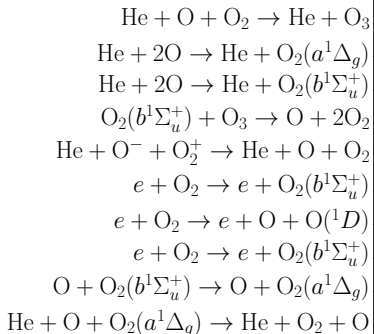
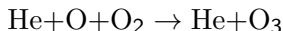
$$\mu_{j,i} = \langle \Delta n_{j,i} / \Delta k_i \rangle$$

# Sensitivity

- Outcome of sensitivity analysis is a “ranking”:

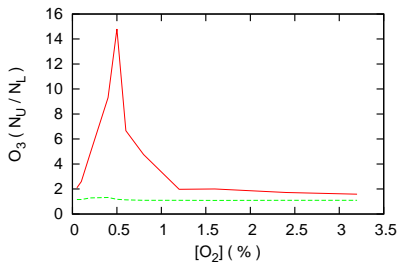
$$\mu_{j,i} = \langle \Delta n_{j,i} / \Delta k_i \rangle$$

- E.g.* Uncertainty in O<sub>3</sub> density is dominated by uncertainty in the rate constant for:



# Optimisation

- 9 sensitive reactions contribute most of the uncertainty
- Optimisation takes advantage of this sensitivity

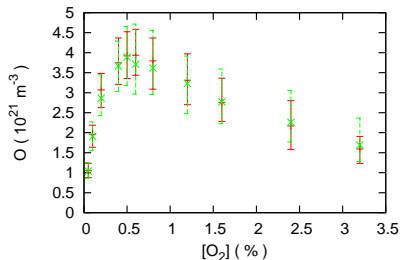
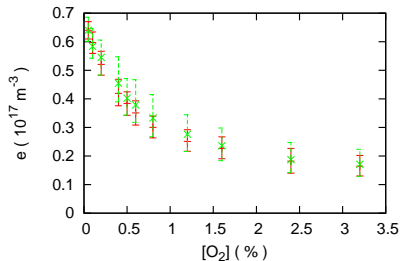


## Did we need 373 reactions?

- 1 Limit the conditions of interest:

$$0.1 \text{ W} < P < 10 \text{ W}$$
$$0.05 \% \leq [\text{O}_2]/[\text{He}] \leq 3 \%$$

- 2 Limit the species of interest:  
 $\text{O}$ ,  $\text{O}_2(a^1\Delta_g)$ ,  $\text{O}_3$
- 3 Discard all reactions contributing  $< 5 \%$  to any time derivative
- 4  $N : 25 \rightarrow 12$ ,  $M : 373 \rightarrow 51$



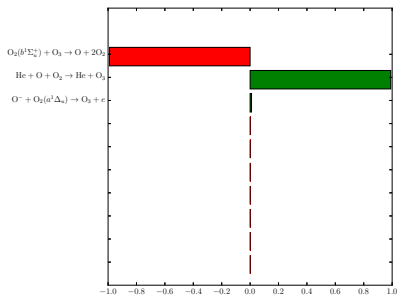
# Procedure

- Clarity of purpose:
  - ① What species densities do we aim to predict?
  - ② Under what conditions?
- Model construction:
  - ① Gather data (provenance!)
  - ② Model reduction  $\Rightarrow$  Selection of relevant processes
  - ③ Sensitivity analysis  $\Rightarrow$  Identification of problematic data
- Validation:
  - ① Relevant species densities measured
  - ② Critical comparison of model and experiment
  - ③ Optimisation

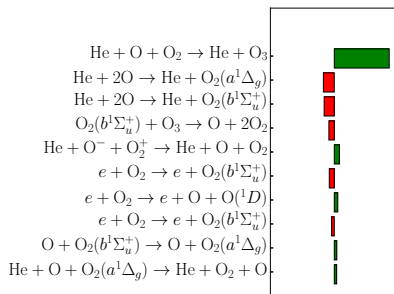
# Dominant Pathways

- We can ask: Which reactions dominantly control species densities?
- A separate question from sensitivity analysis
- Pumpkin is a useful tool

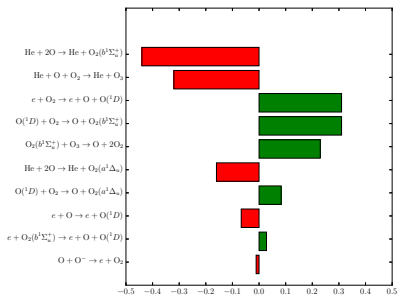
## Dominant pathways



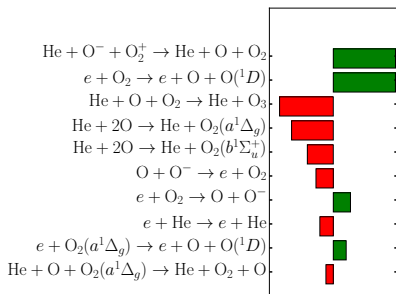
## Sensitivity



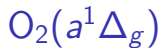
## Dominant Pathways



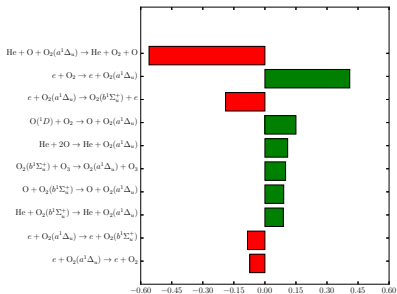
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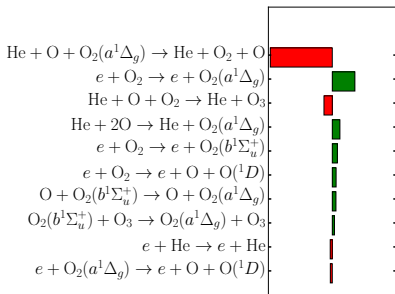




## Dominant pathways

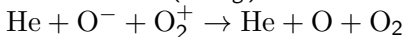
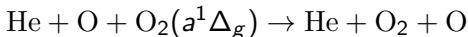
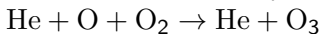
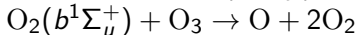
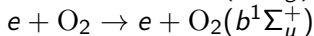
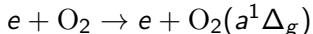
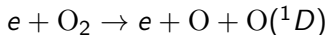


## Sensitivity



## Sensitive Reactions

- Of 373 original reactions, only 9 contribute more than 10 % to the uncertainty of any species of interest:



# Conclusions

- A reaction mechanism is the outcome of a big effort:
  - ① Clear aims
  - ② Curated data with provenance
  - ③ Sensitivity analysis and model reduction
  - ④ Validation
  - ⑤ Optimisation
- The unit of consideration should be a “reaction mechanism” (and not individual rate constants)
- A “reaction mechanism” can often be drastically reduced (important for multi-dimensional models)